

Shape Memory Acrylate Polymers Enabled by Radiation Crosslinking

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Shape Memory Polymers (SMPs)

- Shape memory polymers (SMPs): smart materials which can recover from the temporary shape to the permanent shape upon an external stimulus.
- Two types:
 - Thermoplastic SMP: polyurethane
 - Thermosetting SMP: chemically crosslinked polymers
- SMP manufacturing
 - Thermoplastic SMP: conventional processing
 - Thermosetting SMP: reactive processing
 - OR post-manufacturing processing ionizing radiation



Ionizing radiation

- Two competitive processes during the exposure to radiation
 - Chain scissioning
 - Crosslinking
- Charlesby–Pinner equation

$$s + s^{\frac{1}{2}} = \frac{p_0}{q_0} + \frac{1}{q_0 \mu_1 d}$$

- s: sol fraction
- p₀: degradation density
- q₀: crosslinking density
- d: radiation dose
- $-\mu_1$: starting molecular weight

Earlier work on radiation on acrylate polymers

Previously, we have studied several acrylate polymer systems ۲



Figure 1. Gel fraction as a function of radiation dose for PMA blended with increasing concentrations of a) TAIC; b) TMPTA.

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Voit, W.; Ware, Y.; Gall, K. Polymer 2010, 51(15), pp. 3551–3559

300

Earlier work on radiation on acrylate polymers

3. MA-IBoA copolymer and PLA:

To study radiation temperature effects



Shaffer, S.; Yang, K.; Vargas, J.; Di Prima, M.A.; Voit, W. Polymer 2014, 55(23), pp. 5969–5979



Continuing work

• Acrylate copolymer synthesis

	BuA	IBoA
Sample #1 composition (wt%)	25	75
Sample #2 composition (wt%)	33	67



n-Butyl acrylate (BuA)



Isobornyl acrylate (IBoA)

Thermogravemetric Analysis (TGA)



Fig.4 TGA curves of copolymers of 25% BuA and 75% IBoA

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Fig.5 TGA curves of copolymers of 33% BuA and 67% IBoA

- No obvious difference in thermal decomposition temperature before and after irradiation
- Radiation did not change samples' thermal decomposition temperatures but improved the homogeneity.
- Crosslinking density may be low, if any.



Solvent test

- All samples were soaked in chloroform at room temperature for a week.
- None of the irradiated samples dissolved all swelled in chloroform
- All irradiated samples were lightly crosslinked

Sample composition	Dose (kGy)	Swelling rate in chloroform (%)
25%BuA-75%IBoA	50	_
	100	2118.52
	200	1571.39
33%BuA-67%IBoA	50	_
	100	1571.39
	200	2078.00

*Samples irradiated at 50 kGy: swelled samples difficult to separate out from solvent

Differential Scanning Calorimetry (DSC)



Fig.6 DSC curves of copolymers of 25% BuA and 75% IBoA

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Fig.7 DSC curves of copolymers of 33% BuA and 67% IBoA

• All irradiated samples showed higher glass transition temperatures

Dynamic Mechanical Analysis (DMA)



• In DMA, samples irradiated at 50 kGy showed the highest T_g

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 All rubbery modulus of irradiated samples dropped lower than 100 kPa – lightly crosslinked



Conclusion and future work

- BuA-IBoA copolymers were crosslinked when exposed to e-beam radiation at 50, 100, and 200 kGy, and the crosslinking density was low
- TGA showed no difference in thermal decomposition temperature between unradiated and irradiated samples
- DSC showed all irradiated samples have similarly higher T_g , while DMA showed samples irradiated at 50 kGy have the highest T_g



- Future work:
 - Investigate the specific targets of radiation crosslinking of BuA-IBoA copolymers
 - Employ TGA-MS (mass spectroscopy) to study the side groups of irradiated polymers, in order to further investigate how the radiation at 100 and 200 kGy affect the samples' glass transition temperatures



Reference

- 1. Charlesby, A.; Pinner, S. P Roy Soc A-Math Phy 1959, pp. 367–386
- 2. Voit, W.; Ware, Y.; Gall, K. Polymer 2010, 51(15), pp. 3551–3559
- 3. Vijayabaskar, V.; Bhattacharya, S.; Bhowmick, AK. Radiat Phys Chem 2004, 71(5), pp. 1045–1048
- 4. Shultz, A.R.; J Polym Sci 1959, 35(129) pp. 369–379
- 5. Shaffer, S.; Yang, K.; Vargas, J.; Di Prima, M.A.; Voit, W. Polymer 2014, 55(23), pp. 5969–5979
- 6. Burlant, W.; Hinsch, J.; Taylor C. Polymer Sci 1964, 2, pp. 56-57