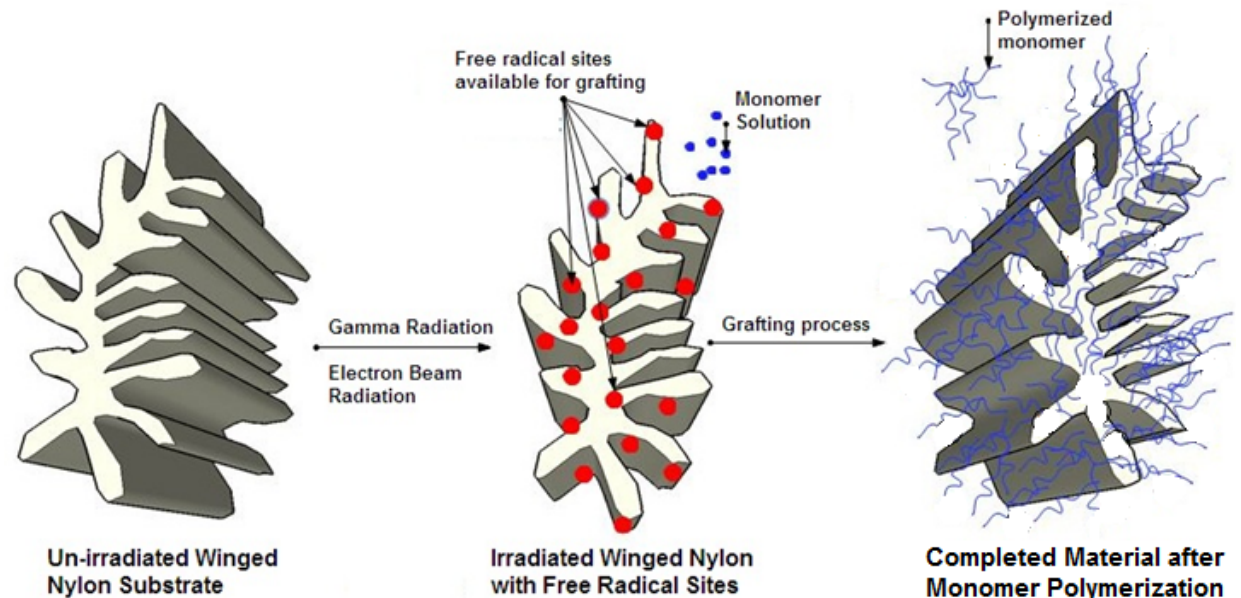


Improvement of the radiation grafting of selective ligands onto polymeric substrates to produce high-capacity adsorbents for harvesting uranium from seawater

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Mohammad Adel-Hadadi²,
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Standards and Technology
(NIST)³



CIRMS 2015 Conference

National Institute of Standards and Technology
Gaithersburg, MD, April 27-29, 2015

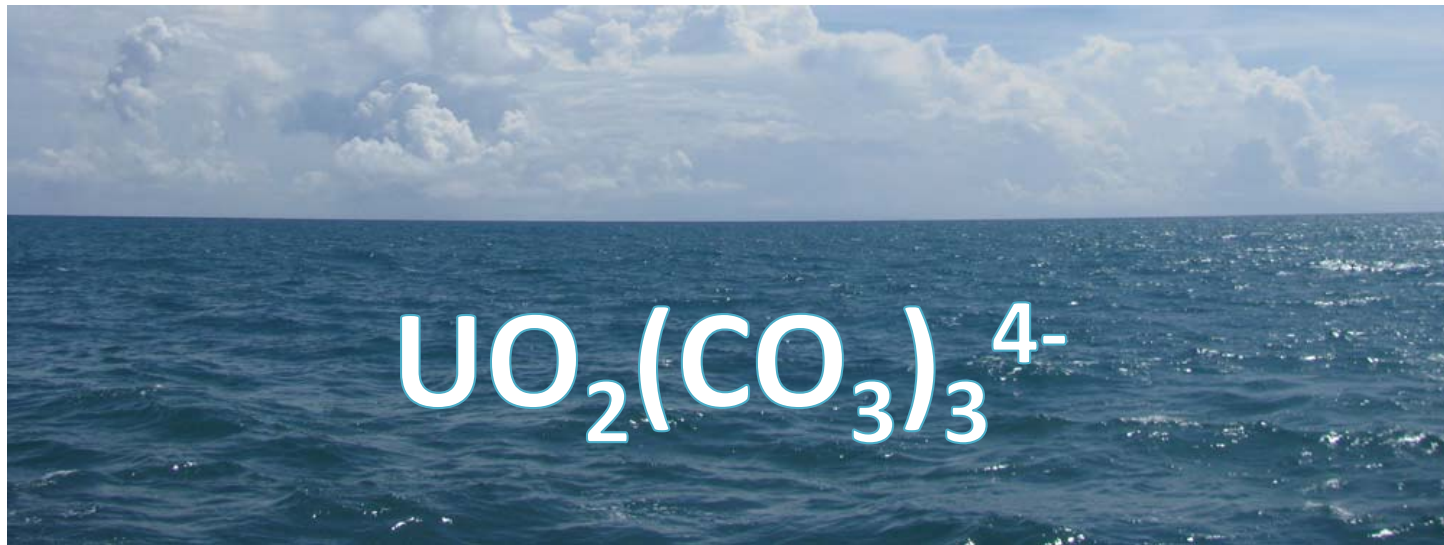


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There exists almost 1000 times more uranium in the world's oceans than is present in terrestrial ores.

However this uranium is:

- **Found at extremely low concentrations (~3.3 ppb)**
- **Amidst the presence of other solutes at much higher concentrations**



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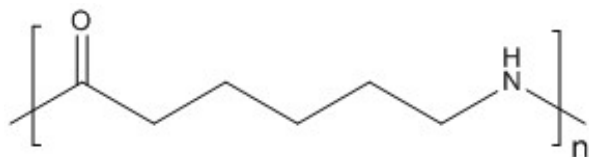
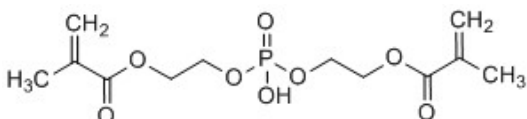
http://www.nefsc.noaa.gov/rcb/photogallery/scenic/photos/ocean4_fullsize.jpg

Kim, J. et al. Recovery of Uranium from Seawater: A Review of Current Status and Future Research Needs. Separation Science and Technology 48, 367–387 (2013).

Our synthetic approach focuses on using radiation to initiate grafting of uranium chelating groups functionalized with vinyl groups.

Materials Selection

- Select ideal adsorbent compounds
- Select ideal substrate materials (polymers)

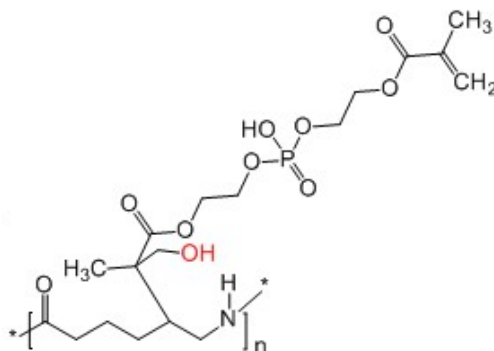


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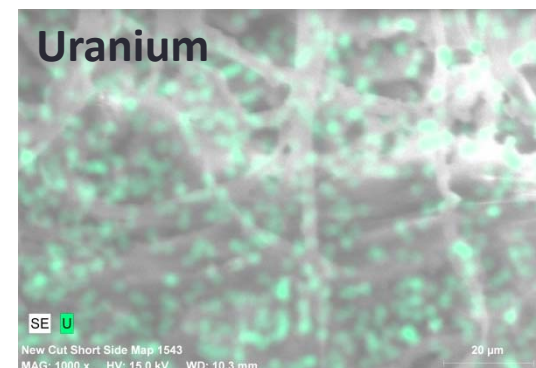
Radiation Grafting

- Single Step synthesis
- Maximize radical concentration for grafting
- Increase length and number of adsorbing groups (grafting density)

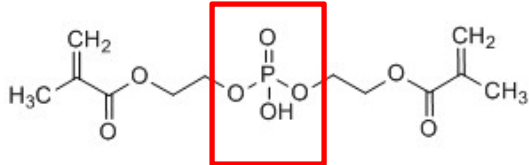
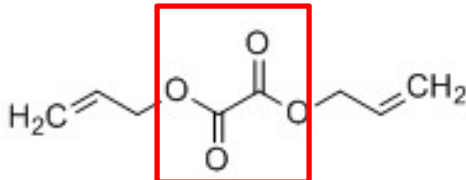
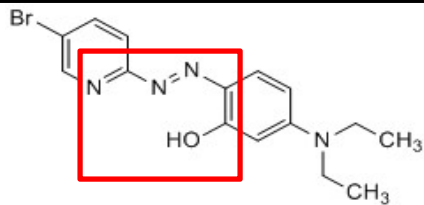


Extraction Testing

- Improve extraction efficiency
- Test scaled-up materials on-site



Prior to grafting experiments, monomers must be chosen and tested for their uranium extraction potential.

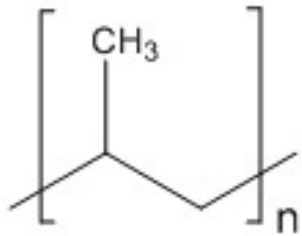
Monomers	Structure
<p>Bis[2-(methacryloyloxy)ethyl] phosphate</p> <p>(B2MP)</p>	
<p>Diallyl Oxalate</p> <p>(DAO)</p>	
<p>2-(5-Bromo-2-pyridylazo)-5-(diethylamino)phenol</p> <p>(Br-PADAP)</p>	



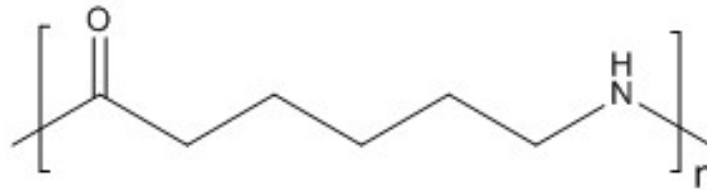
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A number of polymer substrates were selected, tested, and compared.

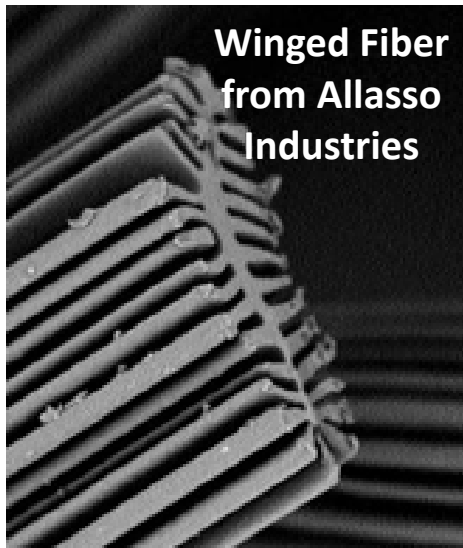
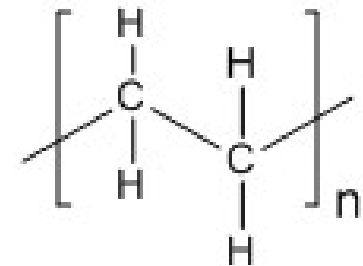
Polypropylene



Nylon 6



Polyethylene



Substrate samples ready for adsorbance testing (in bag)



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TD1

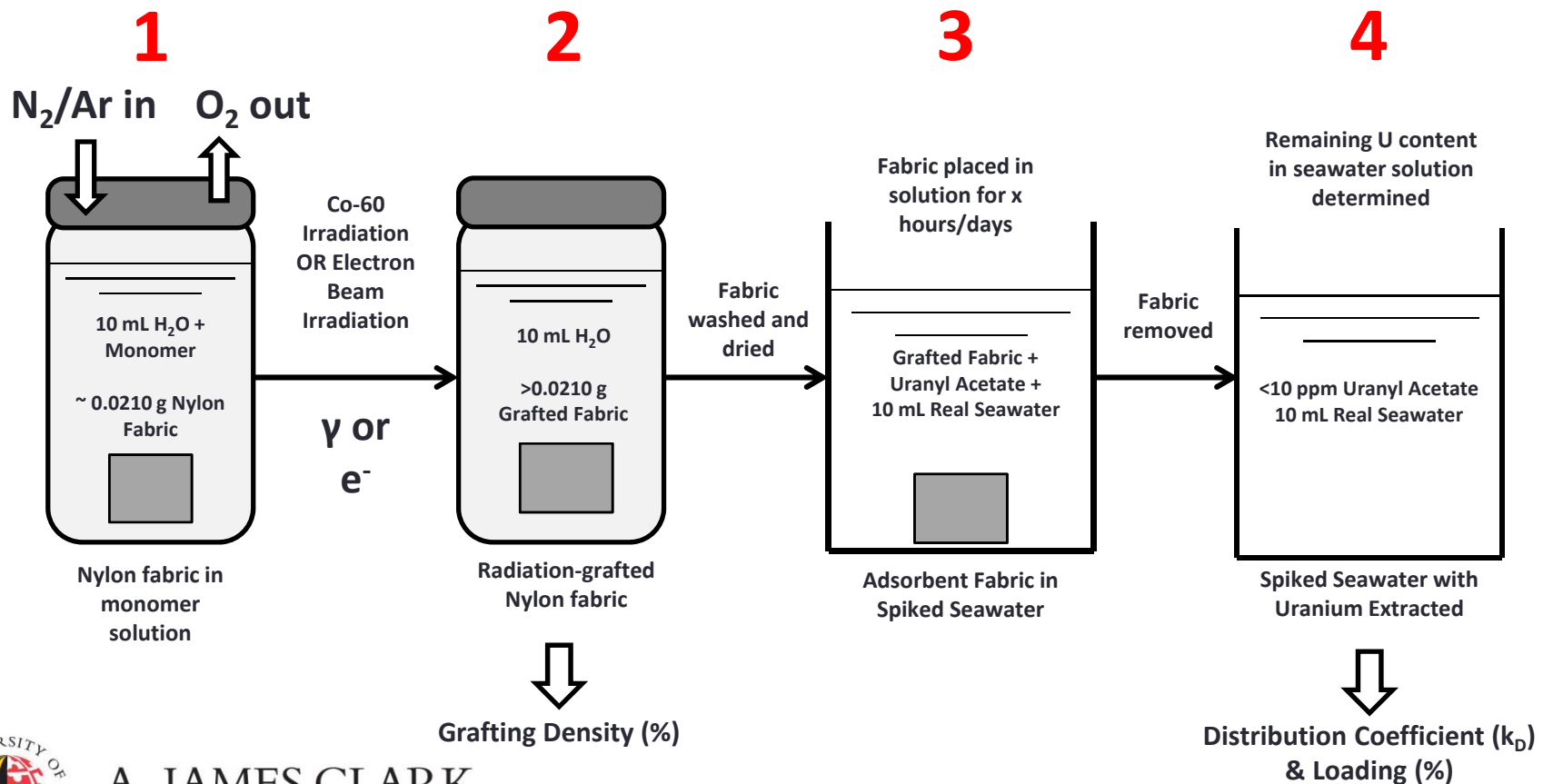
Slide 5

TD1

Mention EPR testing?

Travis, 3/17/2015

The direct grafting method allows for a one-pot synthesis of adsorbent fabrics

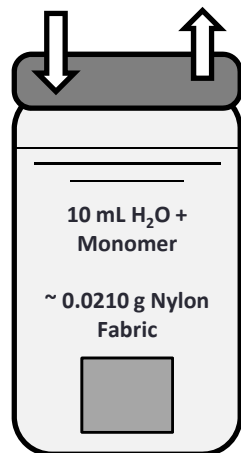


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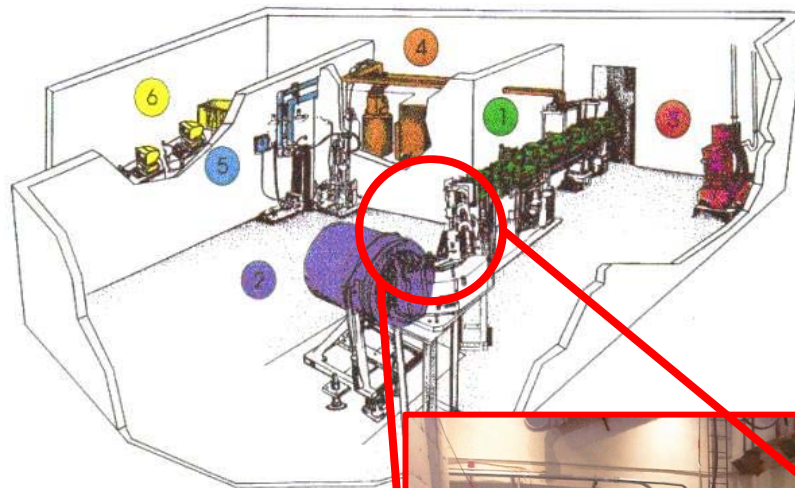
1.) Nylon sample is placed in aqueous solution containing monomer and irradiated

N_2/Ar in O_2 out

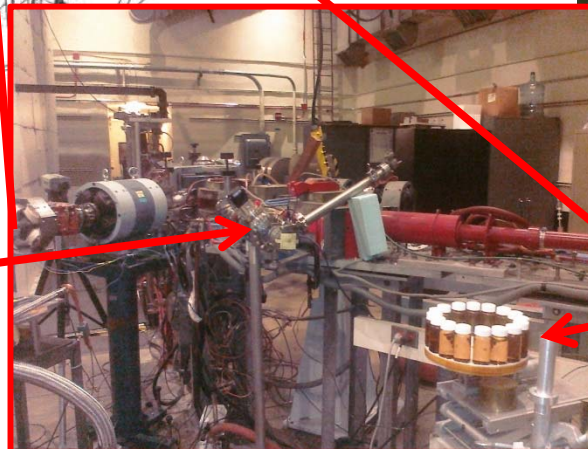


Nylon fabric in monomer solution

Medical-Industrial Radiation Facility, NIST
(Electron beam)



Beam Exit Port



Sample Rotator

Co-60 Irradiator, NIST
(γ irradiator)



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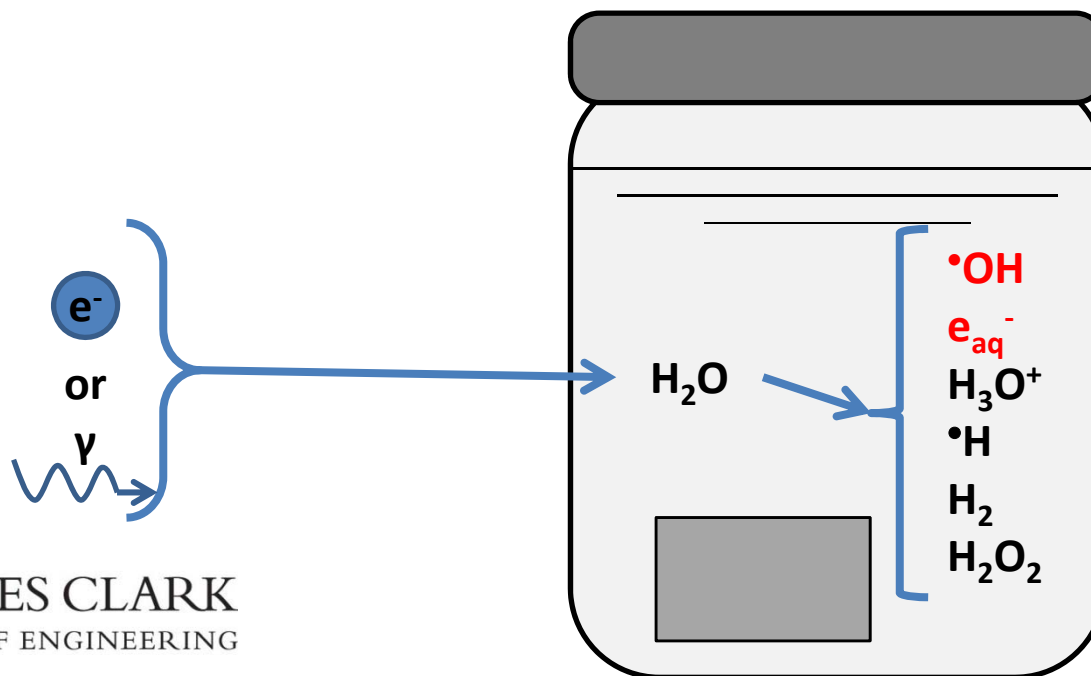
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<http://physics.nist.gov/MajResFac/mirf/mirf-new.jpg>

<http://3dgdos.fjfi.cvut.cz/results/images/gamacel.jpg>

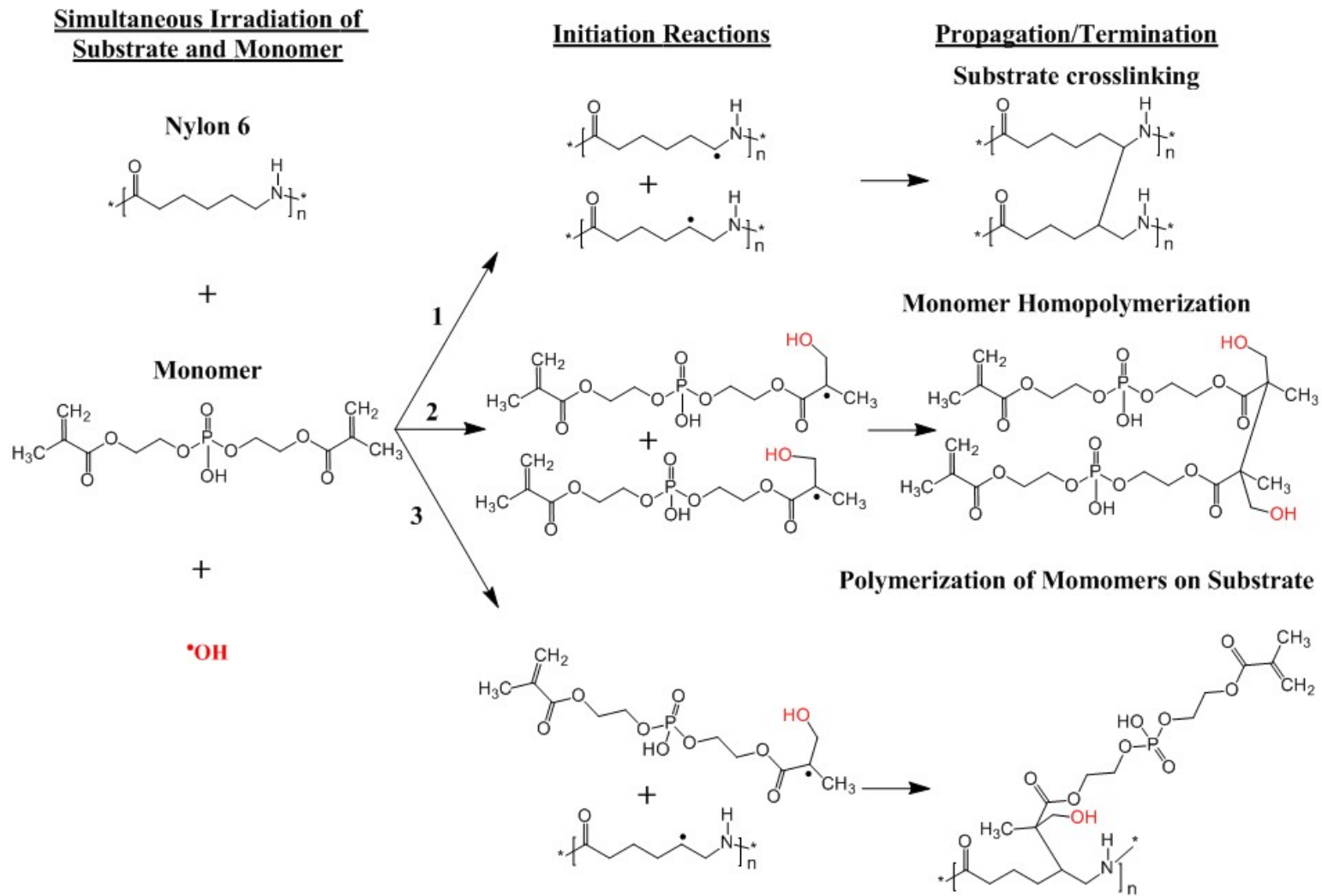
Radiation Chemistry

- Transformation primarily through indirect effects
 - Target molecule: small mass fraction of matrix
 - matrix: generates reactive mobile radicals
- Yields ($\mu\text{mol}\cdot\text{J}^{-1}$) for water irradiated by low LET
 - $G(\cdot\text{OH}) = G(e_{\text{aq}}^-) = G(\text{H}_3\text{O}^+) = 0.28$, $G(\cdot\text{H}) = 0.062$, $G(\text{H}_2) = 0.042$, $G(\text{H}_2\text{O}_2) = 0.082$



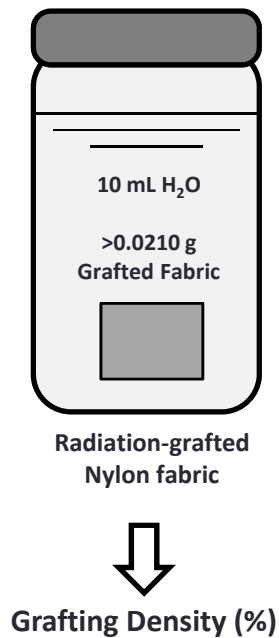
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Initiation, Propagation and Termination Reactions

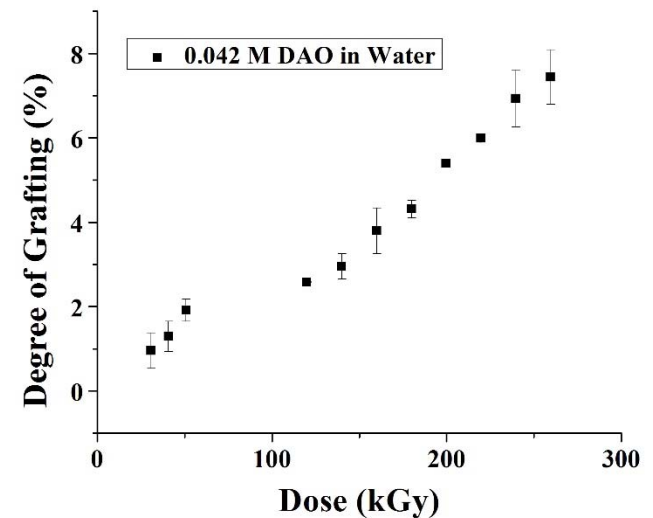
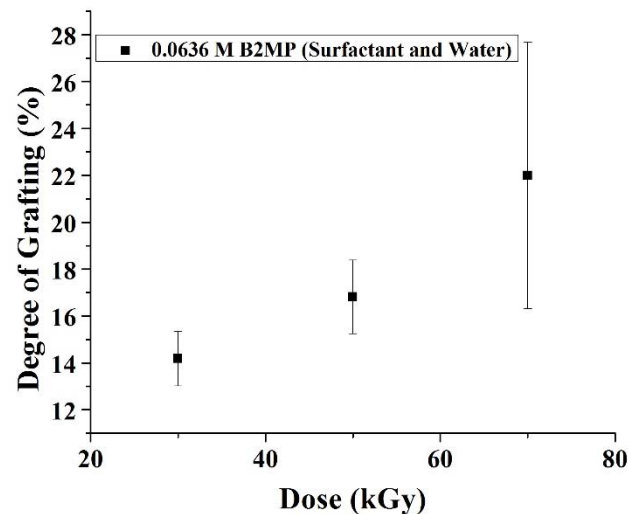


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2.) After irradiation, sample is washed, dried, and massed to determine the grafting density

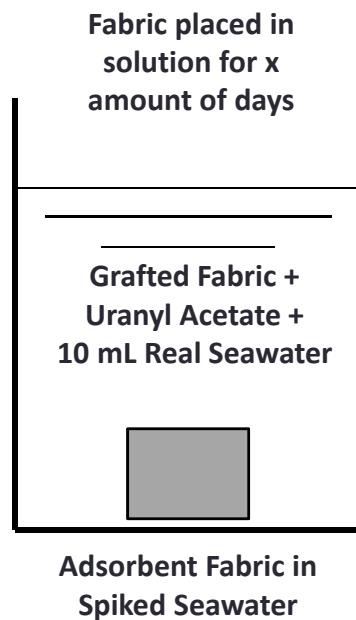


$$\text{Grafting Density} = \frac{(Mass_f - Mass_i)}{Mass_i} \times 100$$



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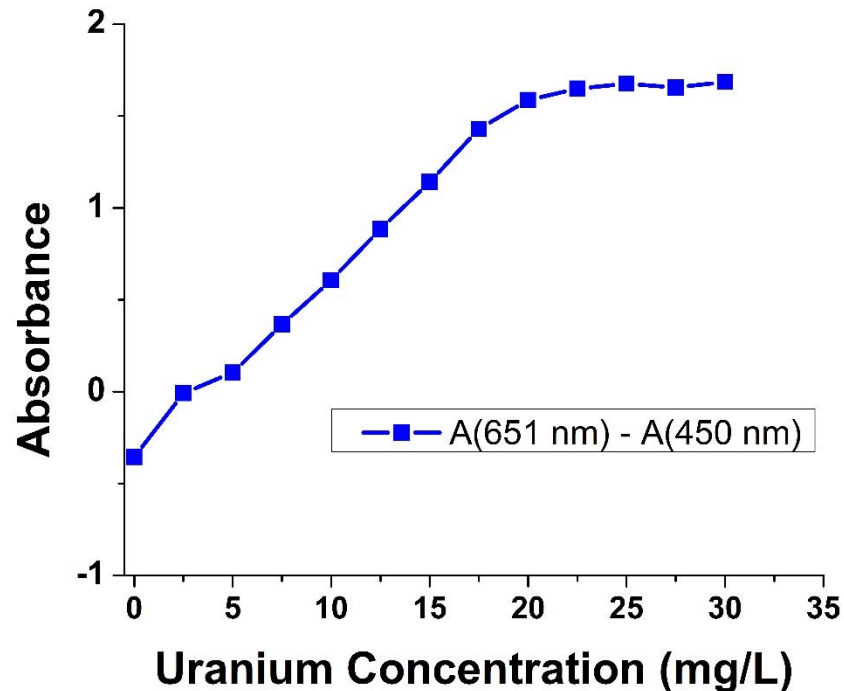
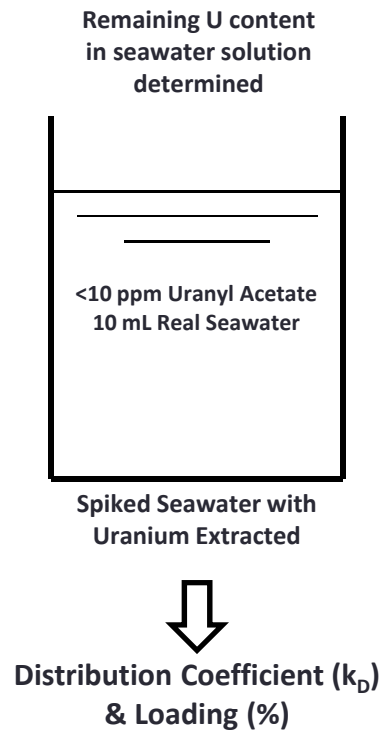
3.) Grafted fabric is tested for its loading capacity by exposing it to aqueous uranium



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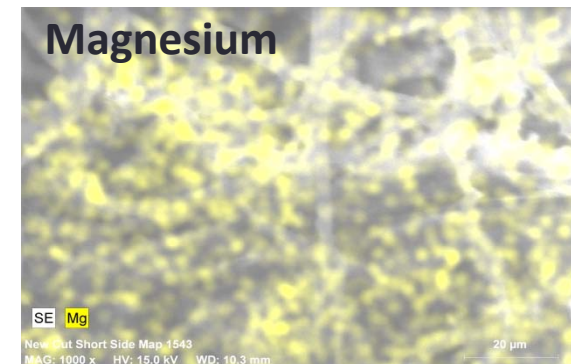
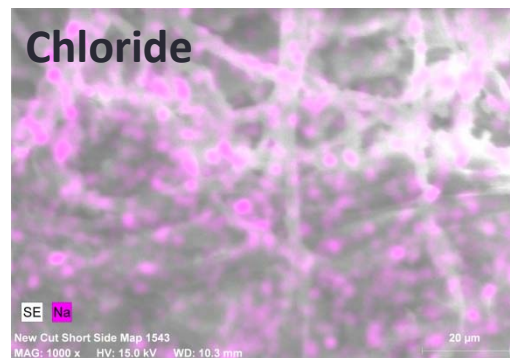
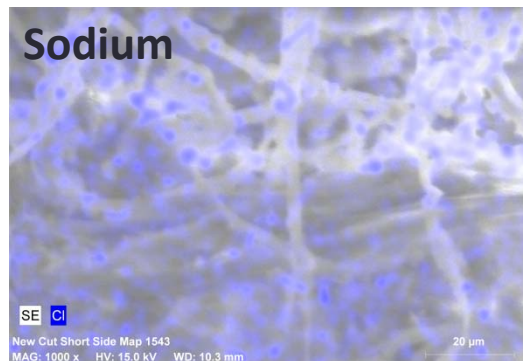
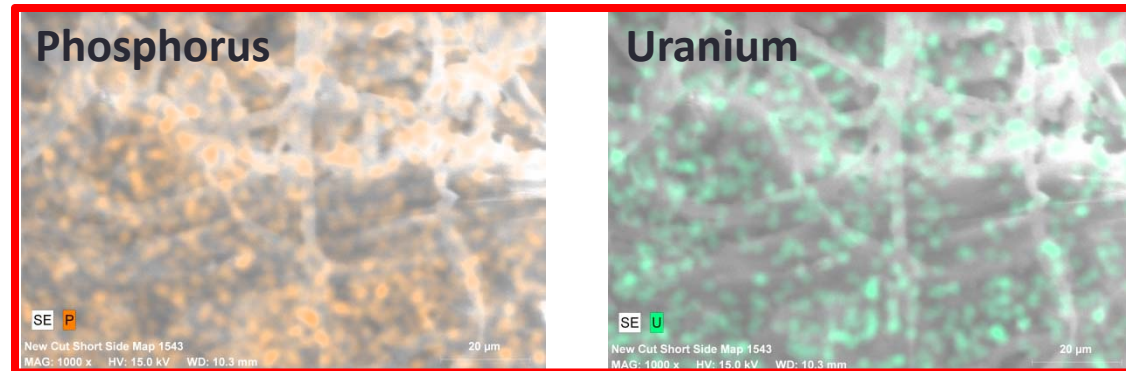
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4.) The amount of uranium remaining in solution is used to determine the distribution coefficient and loading capacity



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EDS studies of the fabric reveal high uniformity of the distribution of adsorption sites for uranium.



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B2MP-grafted Winged Nylon exhibits much higher selectivity for U over Na as compared to Tamada's adsorbent

Amidoxime-grafted fibrous polyethylene (Tamada, 2009)	B2MP-grafted Winged Nylon 6 (present work)
Concentrations on adsorbent: Na: 618.5 $\mu\text{g Na/g adsorbent}$ or 26.9 $\mu\text{mol Na/g adsorbent}$ U: 63.72 $\mu\text{g U/g adsorbent}$ or 0.268 $\mu\text{mol U/g adsorbent}$ U:Na ratio on adsorbent = 0.103 by weight or 0.010 by mole	Elemental analysis of surface by EDS: Na: 1.53 wt. % or 1.04 mol. % U: 7.22 wt. % or 0.47 mol. % U:Na ratio on adsorbent = 4.72 by weight or 0.45 by mole



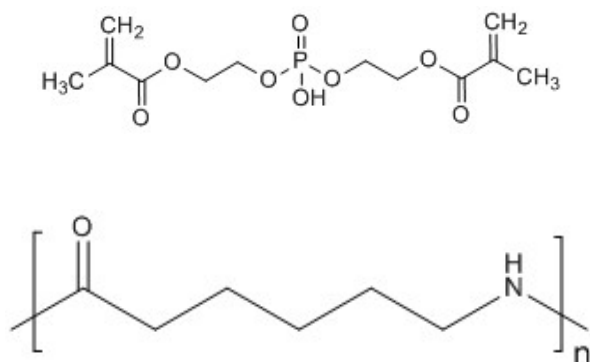
Distinguishing Features of Our Approach

- **One –step synthesis of the absorbent-fabrics.** Elimination of other steps such as amidoximation.
- Performing radiation-induced grafting of the phosphate moiety onto the polymeric fabric under "**green chemistry" conditions**, i.e., in an aqueous medium without need for use of organic solvents.
- **The use of Winged™ polymer fabrics**, in particular, Winged™ nylon 6. This increases the surface area available for grafting.
- **The use of a phosphate, oxalate, and pyridylazo moieties**, consisting of bis[2-(methacryloxy) ethyl] phosphate (B2MP), diallyl oxalate, and Br-PADAP instead of the amidoxime adsorbent groups used by previous research groups.
- **Regeneration with neutral reagent**: The use of a neutral reagent, ammonium oxalate, for multiple cycles of desorption/regeneration with no noticeable degradation of adsorptive capacity

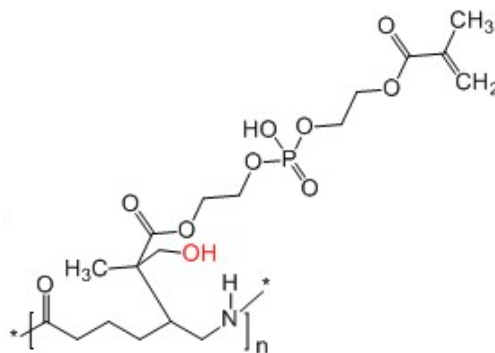


Conclusions

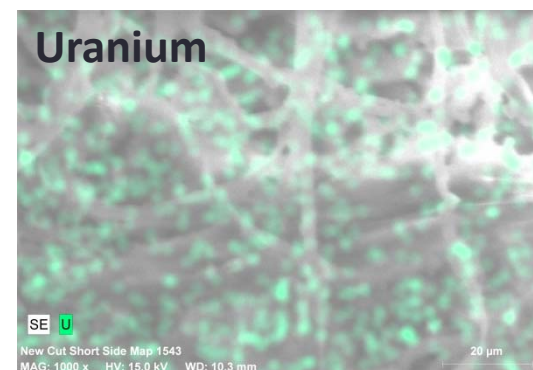
**Able to effectively discover
new potential monomers
for uranium extraction
from seawater**



**Able to effectively graft
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**Able to show that grafted
polymer fabrics can effectively act
as uranium adsorbents in a
seawater environment**



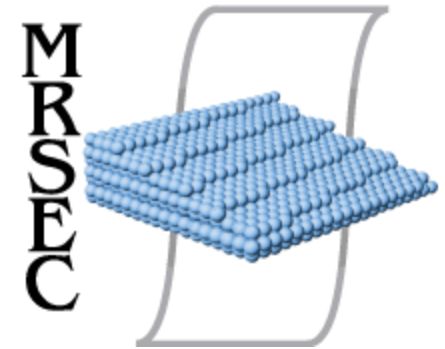
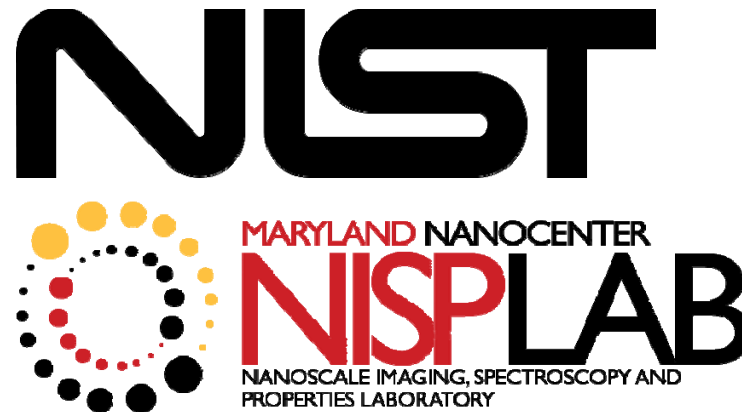
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Acknowledgements

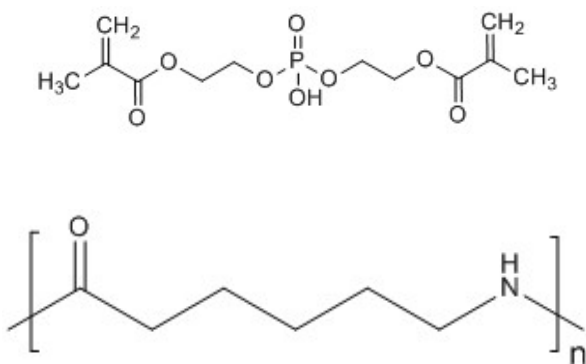
- This project was supported by the Nuclear Energy University Programs through the U.S. Department of Energy
- Maryland NanoCenter and its NispLab
- Dr. Karen Gaskell with the Materials Research Science and Engineering Center at the University of Maryland
- Dr. Fred Bateman and Dr. Lonnie Cumberland and the National Institute of Standards and Technology
- Professor John Ondov
- Tim Mangel with The Laboratory for Biological Ultrastructure at UMD



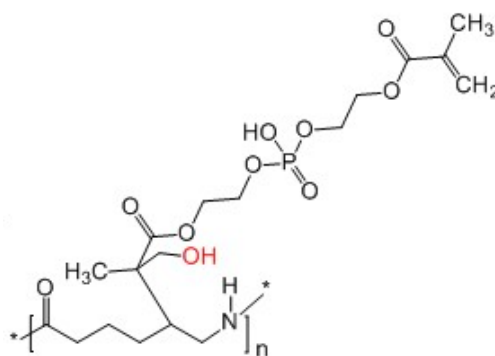
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Conclusions

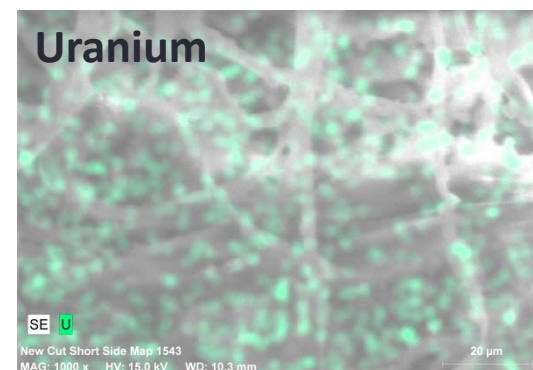
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**Thank you for your time,
any questions?**



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