

# URANIUM ISOTOPE DETECTION USING GAMMA- GAMMA COINCIDENCE COUNTING TECHNIQUES

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# Outline



- Background
- Method
- Setup
- Results
- Future Work

# Background

- Work done through Nuclear Forensics Student Internship Program at Lawrence Livermore National Laboratory in California
- In collaboration with Dr. Tzu-Fang Wang of Lawrence Livermore National Laboratory



# Background

- The ratio of  $^{235}\text{U}$  to  $^{238}\text{U}$  is one of the most important pieces of data in regards to nuclear forensics
- The strongest peak of  $^{235}\text{U}$  is at 186 keV
- There are two main sources of interference at this energy level – Compton continuum and  $^{226}\text{Ra}$

# Background – $^{226}\text{Ra}$

- The strongest gamma peak of  $^{226}\text{Ra}$  is also at 186 keV
- Because  $^{226}\text{Ra}$  is in the decay chain of  $^{238}\text{U}$  it is commonly found in many materials that contain  $^{235}\text{U}$
- Traditionally, these isotopes are separated by finding the amount of  $^{226}\text{Ra}$  from its decay products and subtracting the interference from the 186 keV peak, leaving room for a high amount of uncertainty

# Background – Compton Continuum

- When looking at a spent fuel cell or post-detonation sample, the concentration of fission products in the sample is very high
- Though most of these fission products are very short-lived (seconds or minutes) some of them have much longer half-lives (50+ years)
- These long-lived fission products can have high gamma energies, which contribute to the Compton continuum around the strongest peaks of  $^{235}\text{U}$

# Background - $^{238}\text{U}$

- $^{238}\text{U}$  itself does not have any strong peaks that we can detect
- The daughter of  $^{238}\text{U}$  is  $^{234}\text{Th}$ , which has a 24-day half-life
- The daughter of  $^{234}\text{Th}$  is  $^{234}\text{Pa}$ , which has a 6-hour half-life

# Background - $^{238}\text{U}$

- Because the 1<sup>st</sup> and 2<sup>nd</sup> generation daughters of  $^{238}\text{U}$  have such short half lives, they all come into secular equilibrium fairly quickly
- After about 7-8 months,  $^{234}\text{Pa}$  spontaneously decays just as often as  $^{238}\text{U}$
- $^{234}\text{Pa}$  has a strong gamma peak at 1001 keV, which is outside the Compton continuum, and thus has very little interference associated with it



# Method

- We want to utilize coincident gamma rays in  $^{235}\text{U}$  to isolate it from the interferences affecting its main peak
- There is a strong coincidence in  $^{235}\text{U}$  between the 186 keV peak and the 202 keV peak
- Using proper electronics and timing software, we can find the  $^{235}\text{U}$  concentration using these coincidence events

# Method

- For the  $^{238}\text{U}$  concentration, we can use the 1001 keV peak associated with  $^{234}\text{Pa}$
- By using standards with known concentrations, we can find an equation relating the  $^{235}\text{U}$  coincident events and the  $^{238}\text{U}$  daughter singlet events
- By finding this equation for any particular detector setup, one can determine the isotopic ratio of uranium in any given sample from these coincident and singlet data

# Setup

- At LLNL, the experiment was performed in the Nuclear Counting Facility, which is underneath one story of soil designed to eliminate interference radiation
- In our setup, we had two coaxial HPGe detectors separated by 5 cm, with a Compton suppression system attached to the back of each respective detector

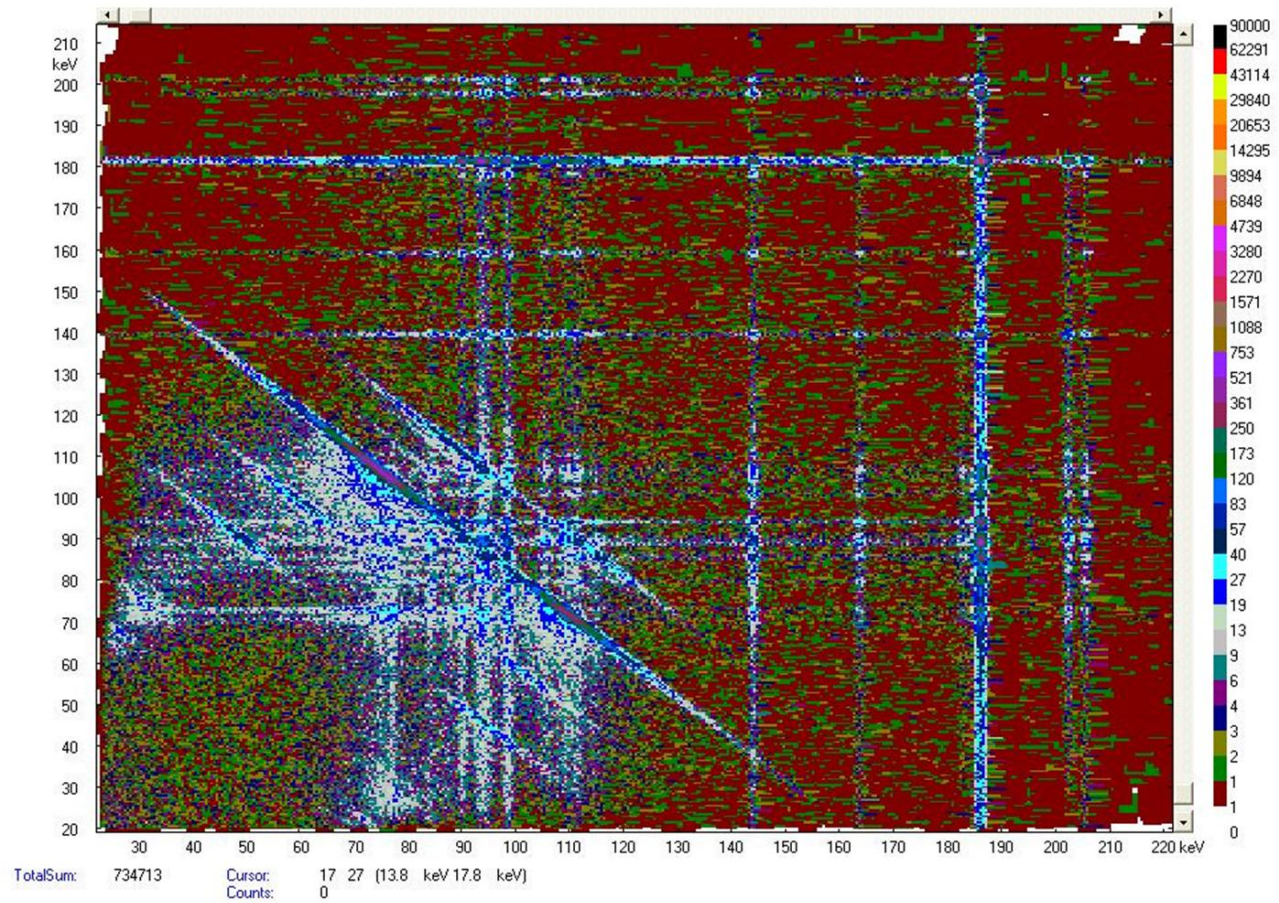
# Setup



# Setup

- The software we used gave single spectrum for each detector, as well as a 2-dimensional plot for the coincidence data
- Using these spectra, the number of counts due to each isotope was determined
- We counted three aged samples consisting of .5%  $^{235}\text{U}$ , 5%  $^{235}\text{U}$ , and 50%  $^{235}\text{U}$  with the remaining portion of the sample being  $^{238}\text{U}$

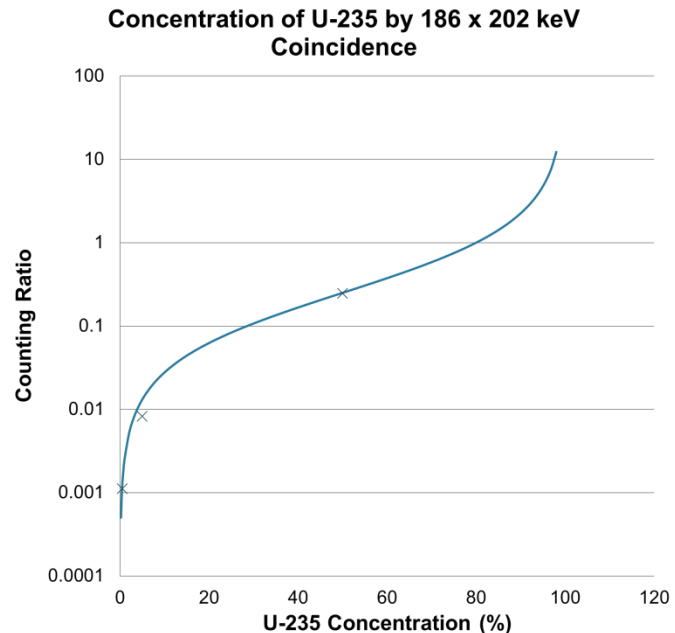
# Results



# Results

- From the data, we were able to obtain an equation relating the ratio of 186-202 keV coincident events to 1001 keV singlet events

U-235 Concentration	186 x 202 Coincidence :: 1001 keV Ratio
.5%	$1.11e-3 \pm 2.00e-4$
5%	$8.19e-3 \pm 2.75e-4$
50%	$2.46e-1 \pm 4.59e-3$



# Future Work

- More data points need to be included in order to give validity to this method, especially at higher  $^{235}\text{U}$  concentrations
- The method needs to be tested under high background conditions, such as those that might be found in a spent fuel cell



# Future Work and Conclusion

- $^{239}\text{Pu}$  also has several coincident gamma rays that can be utilized using a similar method
- Random coincidence subtraction methods could also be useful in gaining accuracy for this method
- This work performed under the auspices of the U.S. Department of Energy by Lawrence Livermore National Laboratory under Contract DE-AC52-07NA27344