



Density Corrections for the Detection of Radioactive Contamination in Food with Gamma Spectrometry

Council on Ionizing Radiation Measurements and Standards
April, 19th 2016

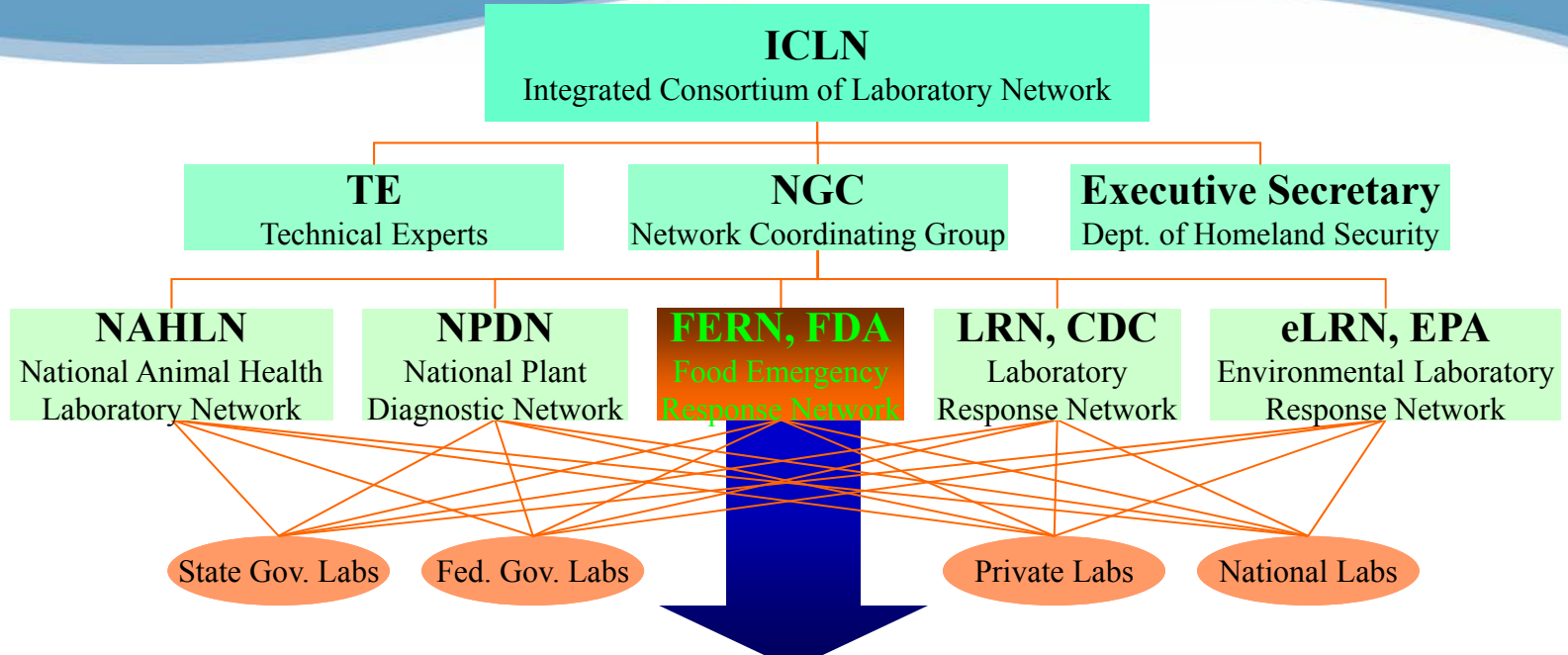
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Food and Drug Administration

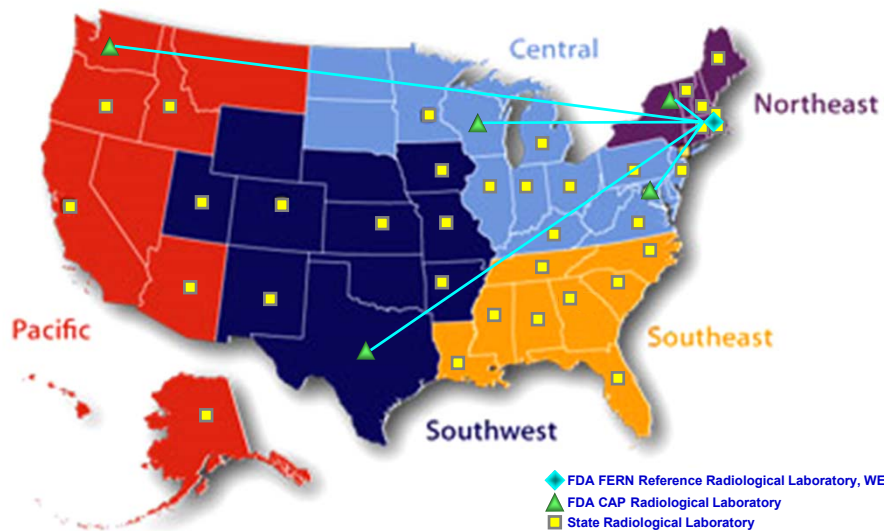


Rapid response, confident risk assessment, and coherent interagency action require:

- A broad analytical capabilities with sufficient surge capacity
- Proficient laboratory analysis with assured data quality
- Data comparability with demonstrated measurement traceability
- Use of analytical methods developed with a harmonized approach



FDA FERN Radiological Laboratory Network



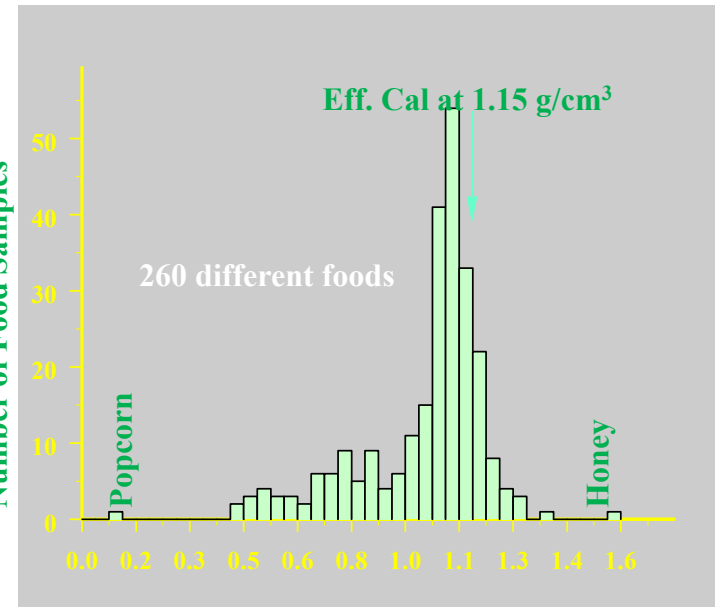
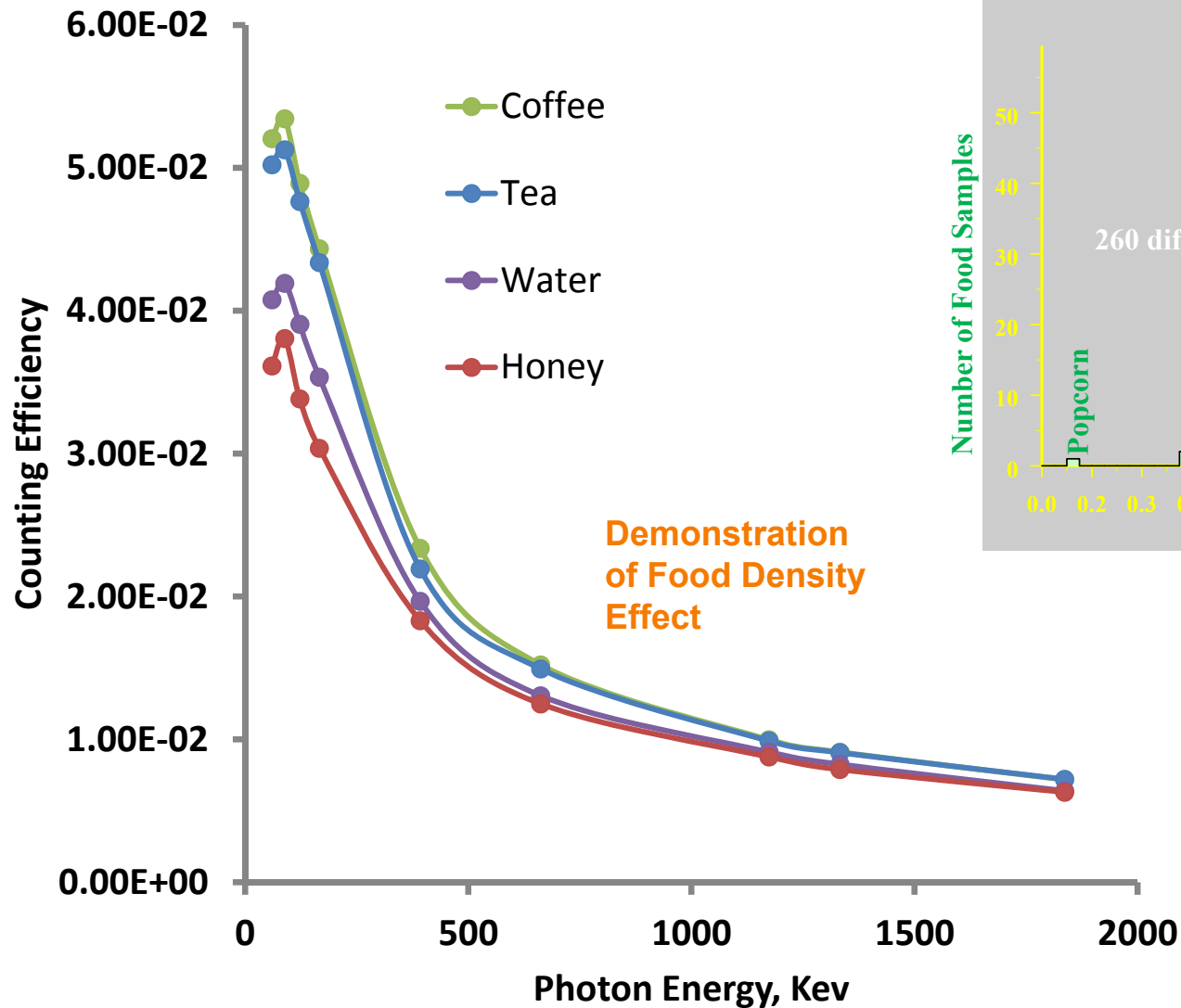


The current limiting factors in food gamma-ray spectrometry analysis are:

- **Precise matrix effects and extent of measurement corrections for analyzing gamma-emitting radionuclides in foods are largely vague**
- **Greater than 50% of mostly concerned gamma-emitting radionuclides show significant coincidence-summing effects**
- **High cost and limited availability of food-based standards force most of FERN network laboratories to practice “one-standard-fit-all” calibration**
- **Significant self-attenuation and coincidence-summing effects are either generally ignored or corrected using different methods**
- **Measurement accuracy and data comparability are compromised due to method discrepancies and generalization**
- **Lack of a simple and practical efficiency correction technique applicable to a broad range of food matrices, counting geometries, and summing radionuclides**



Counting Efficiencies Measured with Different Mixed-Gamma Food Standards



Food Density, g/cm³

Sample density effect in food analysis is real and significant, and self-attenuation effect increases with decrease in photon energy.

Applying one-standard-fit-all calibration will bias measurement results considerably.



Sourceless Efficiency Calibrations

General-purpose Monte Carlo N-Particle code (MCNP) will be used for calculations

Advantages of using calculations

Lower Cost than Source-based Calibration

Purchase of Sources

Replacement of Sources

Disposal of Sources

Easier to match the sample for calibration

Not enough material

Unusual material composition

Each of the three software in this study is comprised of the Monte Carlo N-Particle code, but they all differ by having unique user interfaces and adjustable parameters that provide distinct advantages and disadvantages in comparison to each other.

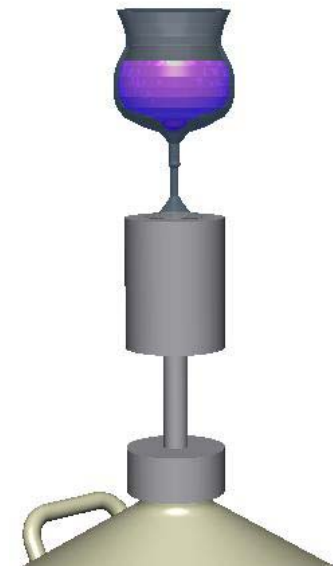
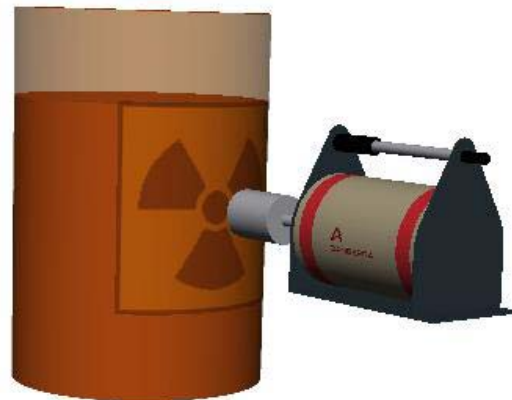
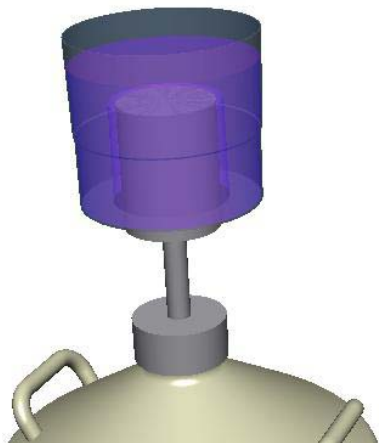


Sourceless Efficiency Calibrations

LABSOCS/ISOCS The LabSOCS (Laboratory SOurceless Calibration Software) mathematical efficiency calibration developed by Canberra.

The difference between LabSOCS and other available mathematical calibration packages is that the detector used to perform activity measurements is initially characterized by the manufacturer.

The software also contains a variety of geometry templates of commercially available containers.

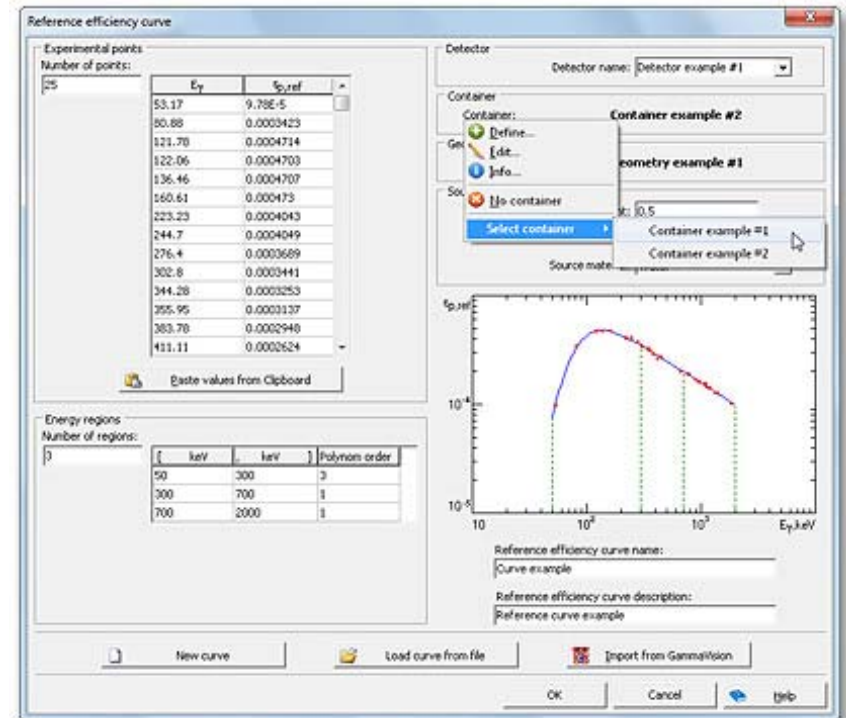


Sourceless Efficiency Calibrations

ANGLE 3 is developed by Ametek and combines advantages of both Monte Carlo calculations and calibrated-source-based methods.

By using both methods, the potential for systematic errors is significantly reduced in the calculations and while the practical limitations of source base corrections is mitigated.

Angle can be used with any HPGE detector, which is advantageous in applying the methods in this study across a range of detectors.

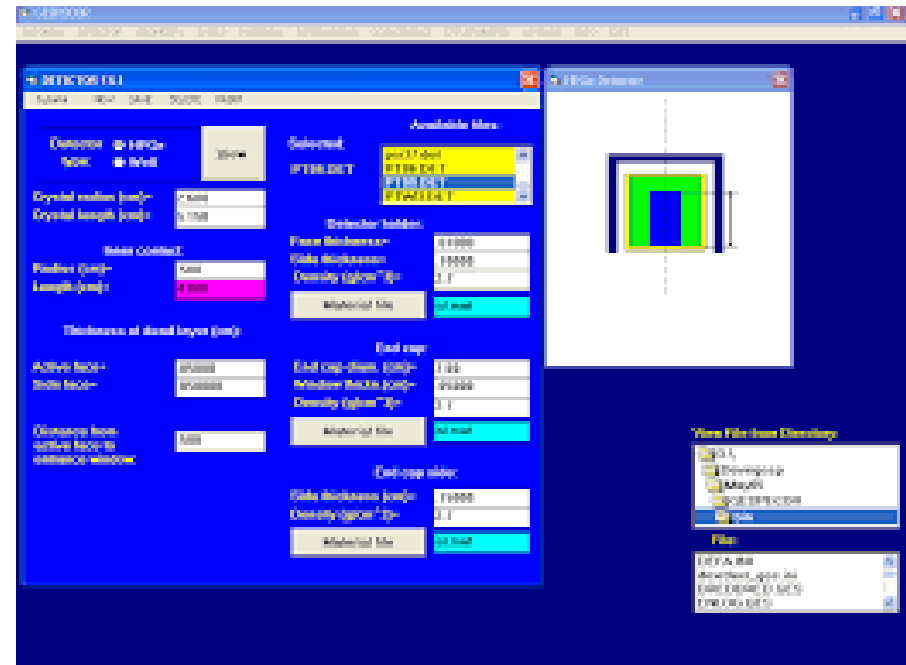


Sourceless Efficiency Calibrations

GESPECOR Germanium Spectroscopy Correction factors is a Monte Carlo based code specially developed by Sima for solving problems in gamma-ray spectrometry measurements.

GESPECOR is able to provide a set of procedures for the efficiency calibration of high resolution gamma ray spectrometers based on hyper pure Ge detectors.

The software is very flexible and allows the user to change many variables, including the detector, shield and geometry all from a friendly user interface.





Computational approach requires descriptive inputs with regard to:

- **Detector Characteristics**
- **Sample Properties**
- **Shielding Material**
- **Counting Geometry Arrangement**

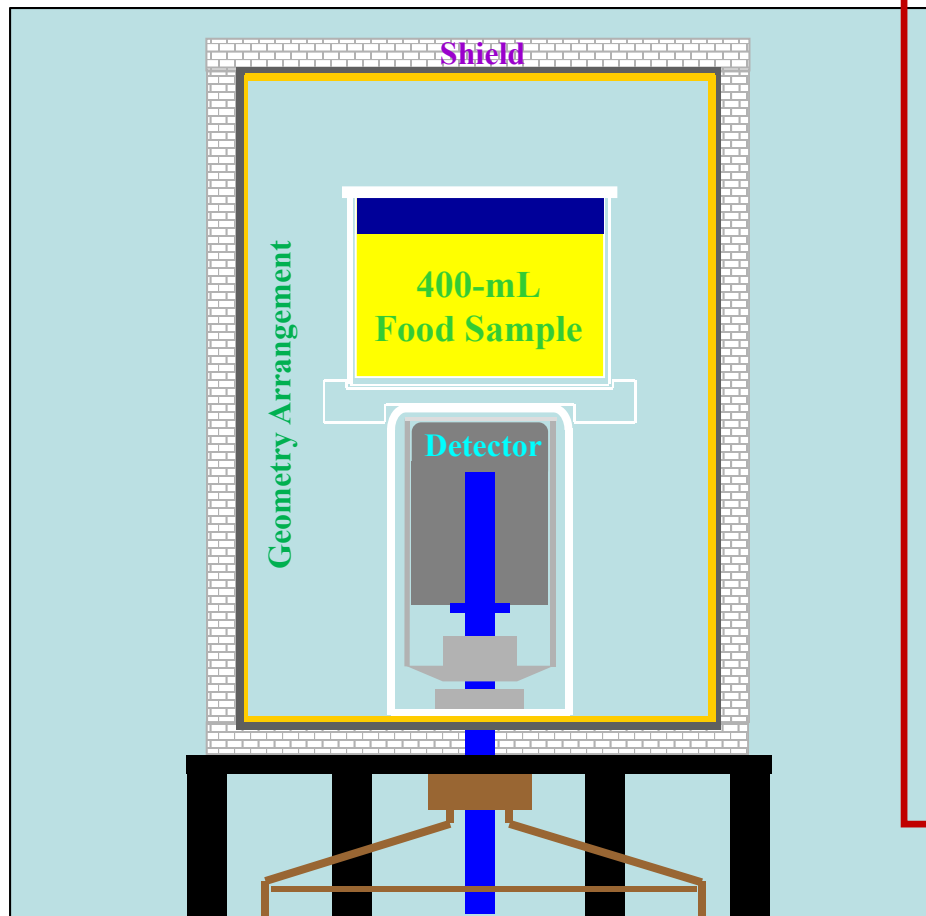
List of Input Variables

- Crystal Radius
- Crystal Length
- Crystal Inner Contact Radius
- Crystal Inner Contact Length
- Crystal Holder Face Thickness
- Crystal Holder Side Thickness
- Crystal Holder Material Density
- Face Dead Layer Thickness
- Side Dead Layer Thickness
- End Cap Diameter
- End Cap Window Thickness
- End Cap Side Thickness
- End Cap Material Density

- Container Radius
- Container Height
- Container Wall Thickness
- Container Material Density
- Sample Composition
- Sample Density

- Shield Inner Radius
- Shield Inner Height
- Shield Outer Radius
- Shield Outer Height
- Shielding Material Density

- End Cap-to-Inner Shield Top Distance
- Active Face-to-Entrance Window Distance



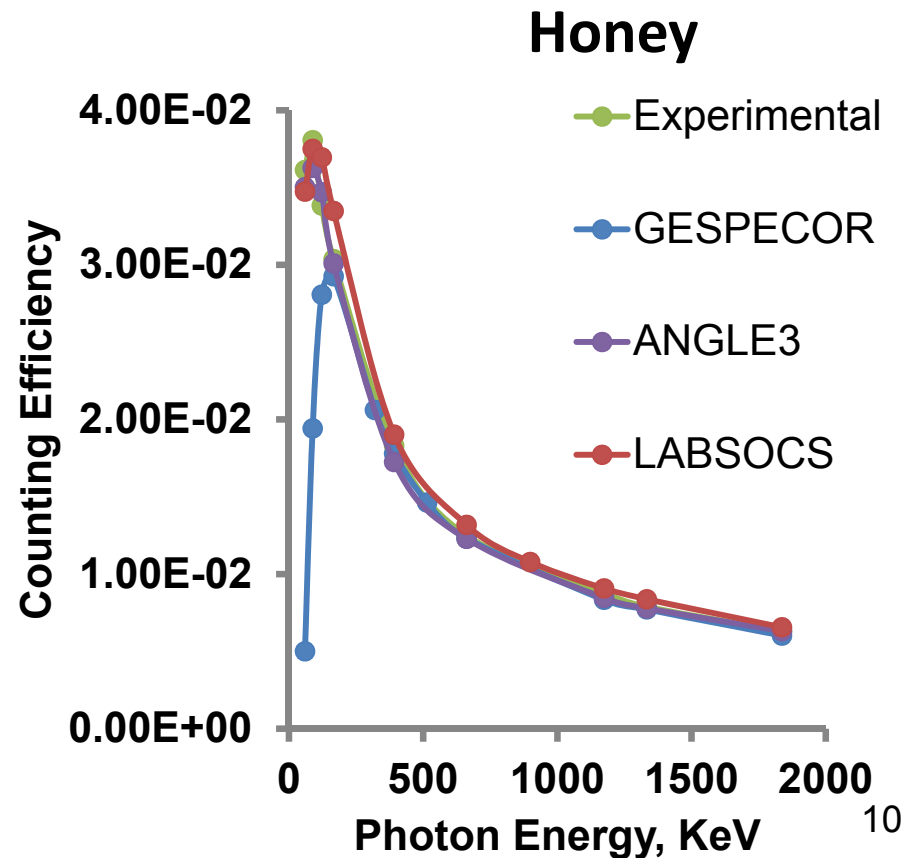
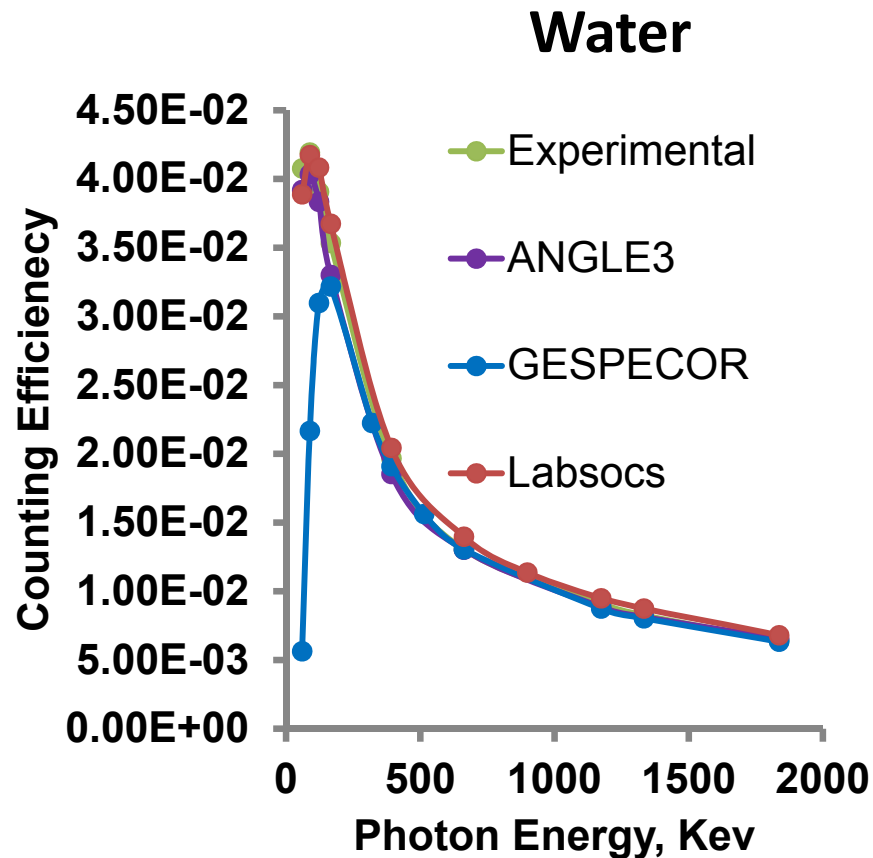


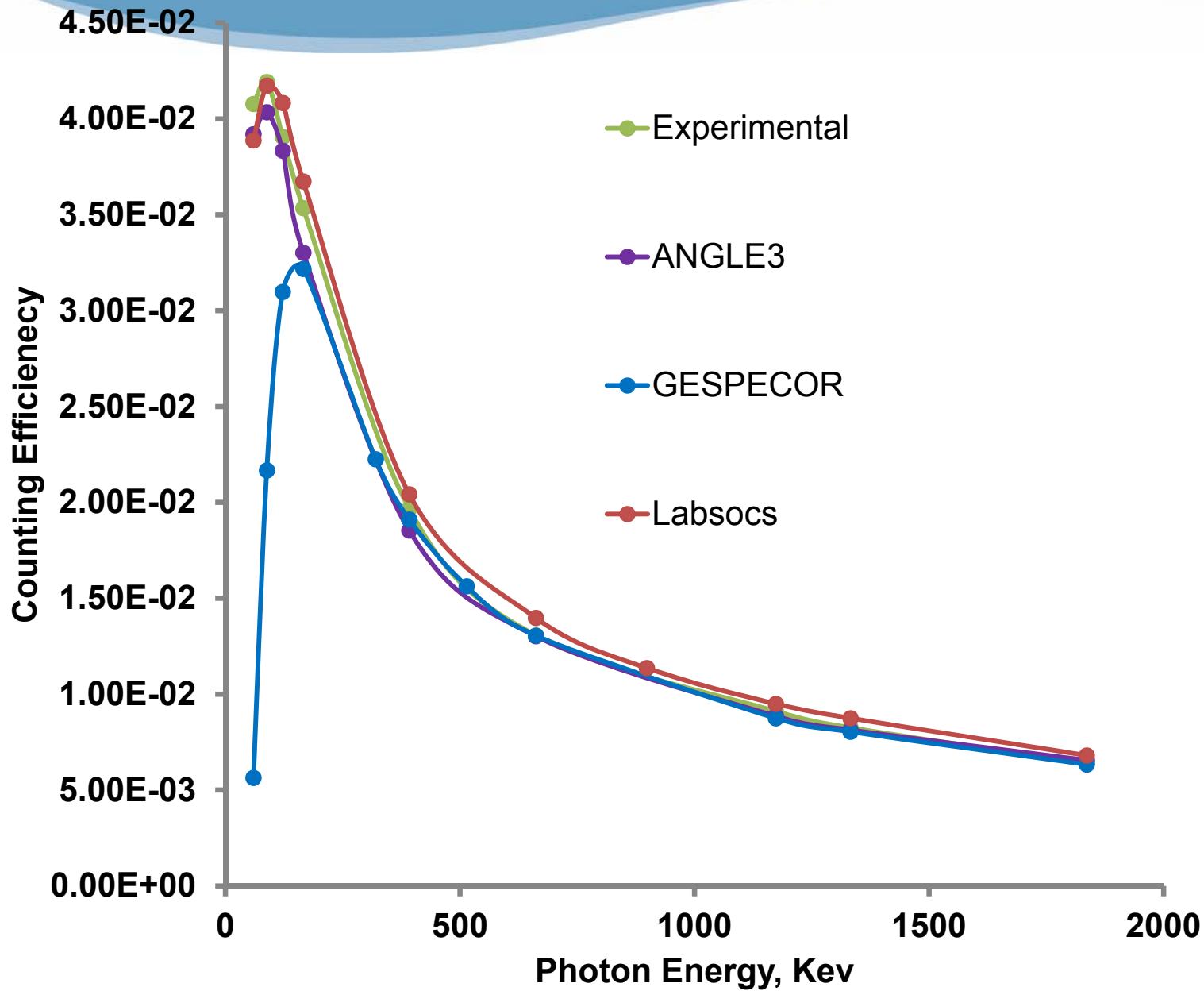
Comparison of Measured Efficiencies with Computed Efficiencies

The **measured efficiency** values are determined using calibration standards

The **LabSOCS efficiency** values are calculated with precise detector parameters

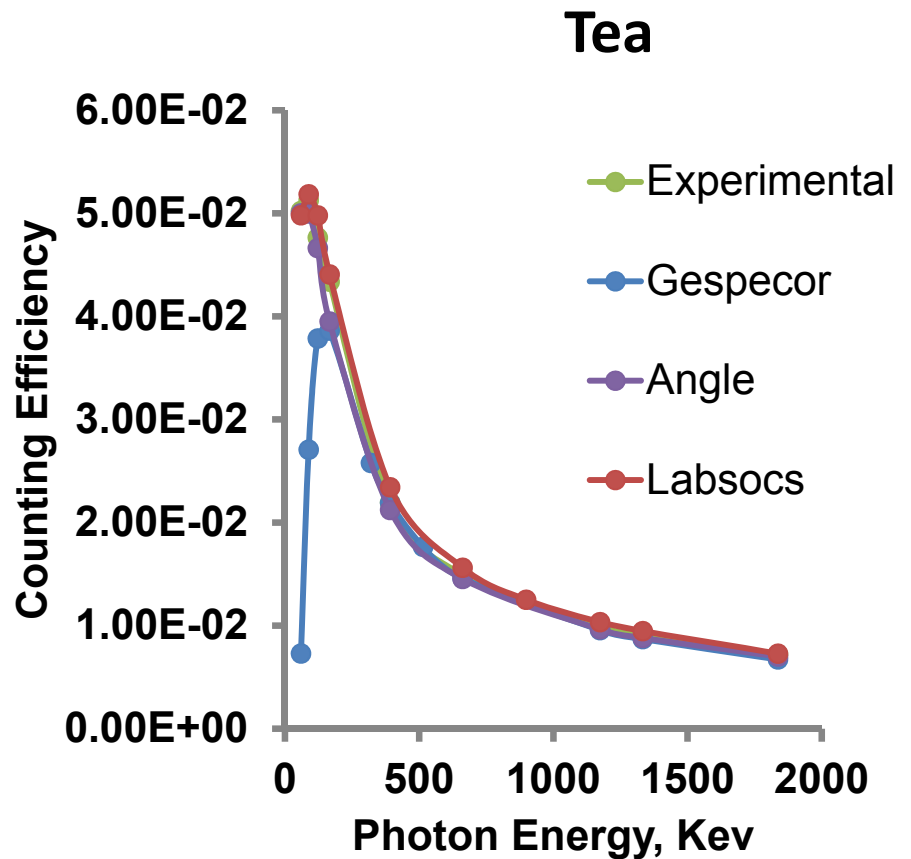
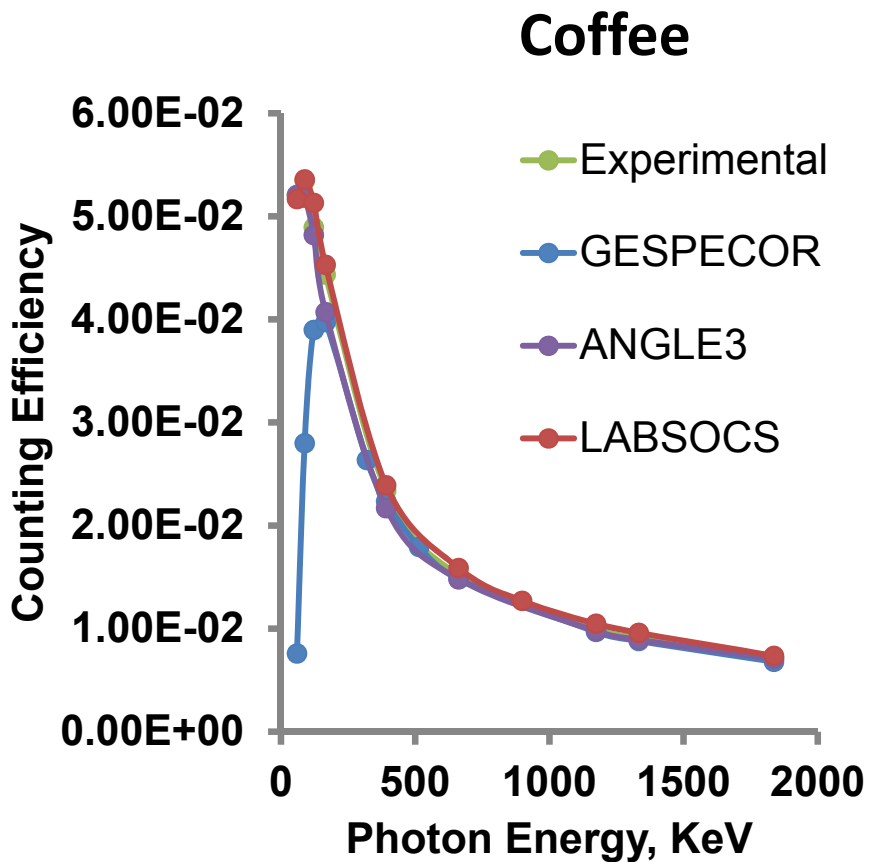
The **GESPECOR** and **ANGLE3** are calculated with generic detector parameters







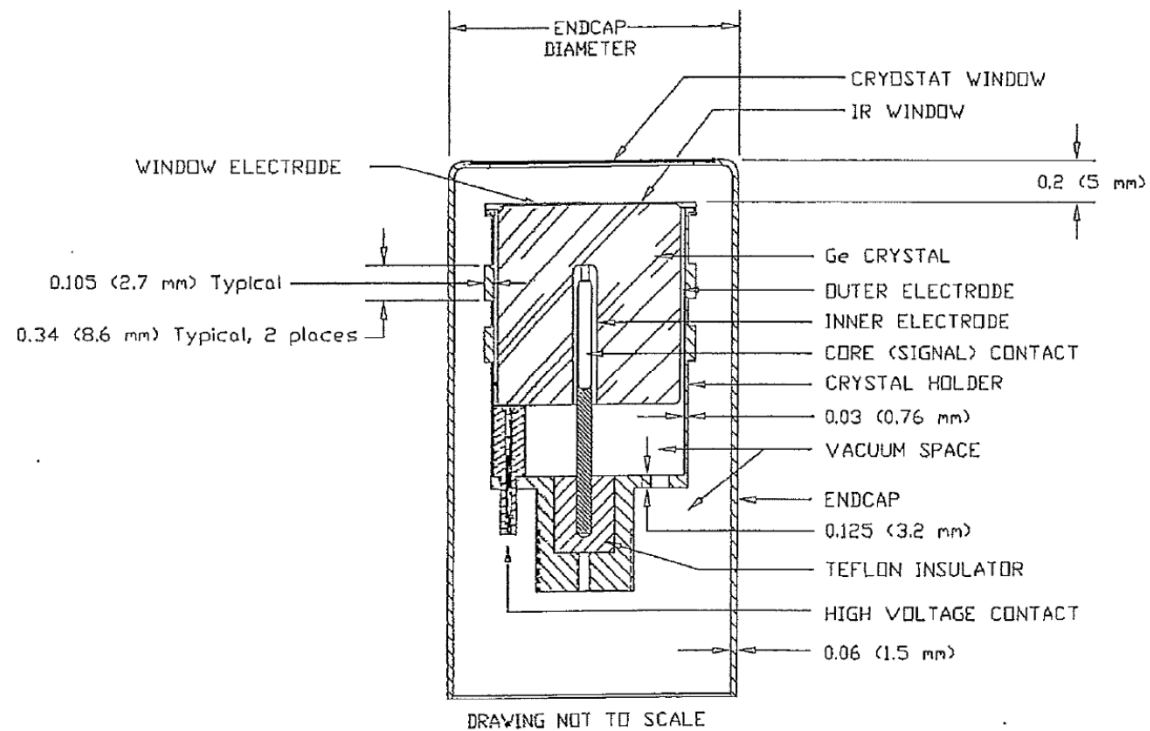
Comparison of Measured Efficiencies with Computed Efficiencies



Comparison of Measured Efficiencies with Computed Efficiencies

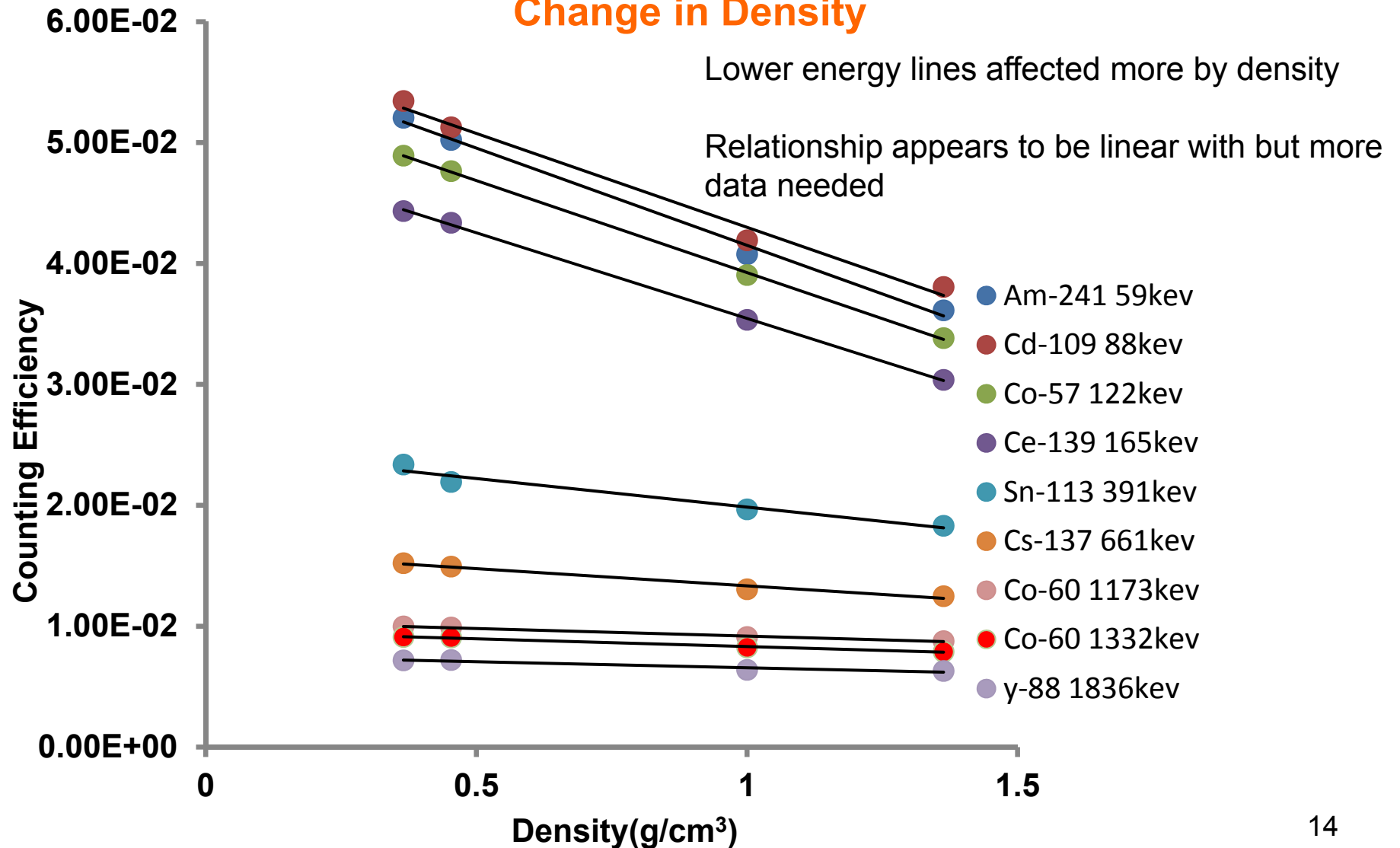
LABSOCS is the best at predicting measured efficiencies

Can detector characterization have unique parameters to improve accuracy?





Comparison of Measured Efficiencies with Change in Density





Comparison of Measured Efficiencies with Change in Density

Food Product	Density	Elemental Analysis
Coffee	0.365	C% 57.00 H% 7.00 O% 35.00 N% 1.00
Tea	0.453	C% 55.59 H% 6.10 O% 35.16 N% 3.02 S% 0.07 P% 0.07
Water	1	H% 11.19 O% 88.81
Honey	1.363	C% 40.00 H% 7.00 O% 53.00

Organic Solvent	Density	Elemental Analysis
Triethylamine	0.7255	C% 71.22 H% 14.94 N% 13.84
Acetone	0.791	C% 62.04 H% 10.41 O% 27.55
DMF	0.944	C% 49.30 H% 9.65 O% 21.89 N% 19.16
Glyme	0.9637	C% 53.31 H% 11.18 O% 35.50
Ethylene Glycol	1.115	C% 38.70 H% 9.74 O% 51.55
Glycerol	1.261	C% 39.13 H% 8.76 O% 52.12

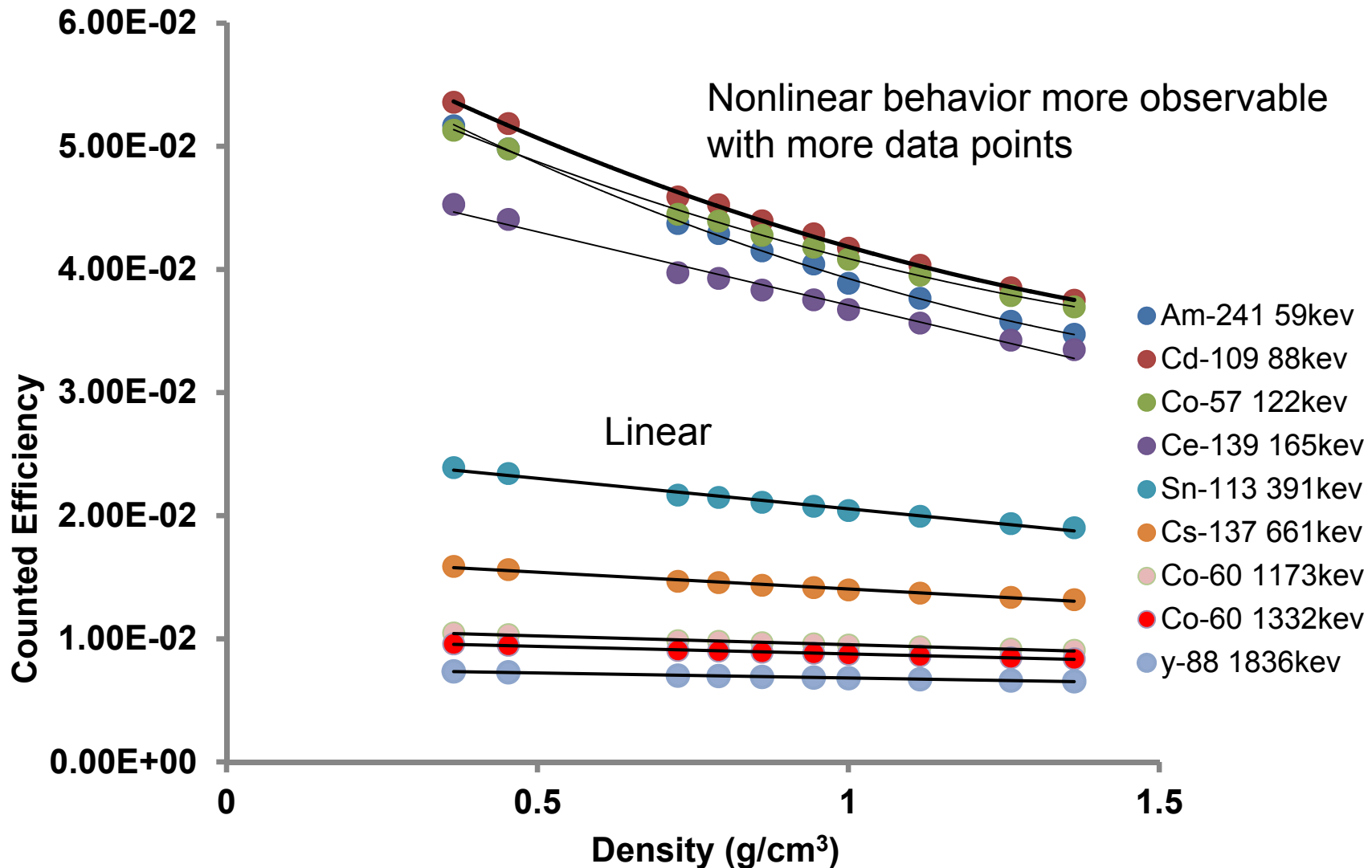
Principle components in food are C, H, O, N

Using organic solutions to mimic food components, while improving range of densities that can be studied

Simulate radioactive solutions to minimize generating waste.



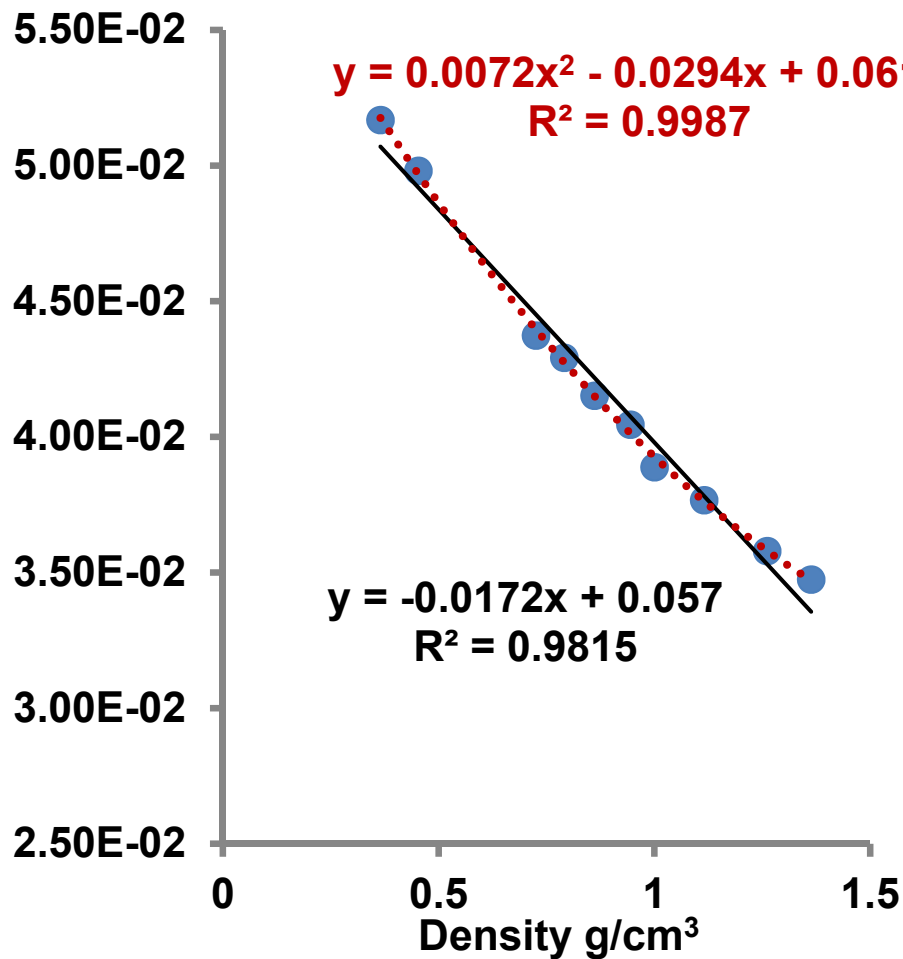
Comparison of Computed Efficiencies with Change in Density



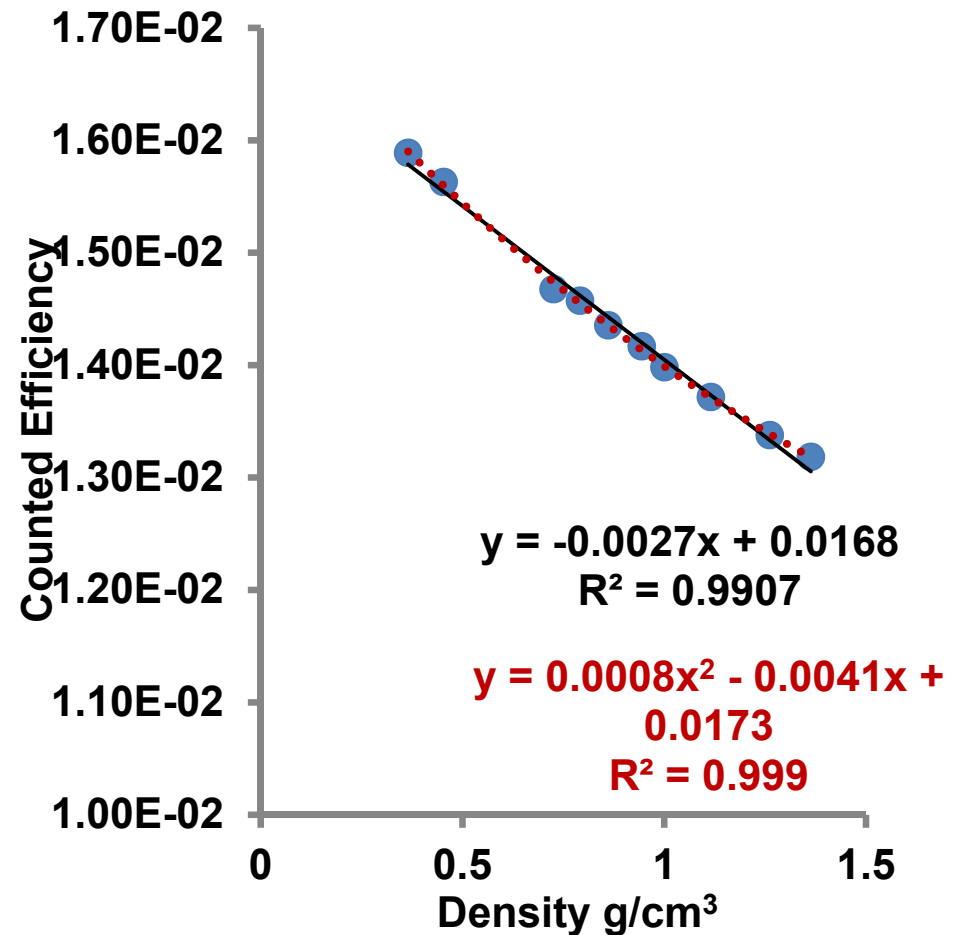


Comparison of Computed Efficiencies with Change in Density

Nonlinear Am-241



Linear Cs-137





Results of Food Spiked With ¹³⁷Cs and ⁶⁰Co

Spiked Activity – The amount of radioactive material added to the food sample.

Measured Activity – The activity determined by measurement of the sample using a water based efficiency

Corrected Activity – The activity determined in the sample using computed efficiencies with density corrections

Food Product	Density	Isotope	Spiked Activity (Bq/g)	Measured Activity (Bq/g)	%error	Corrected Activity (Bq/g)	%error
Water	1	¹³⁷ Cs	0.64	0.66	3.29	0.62	2.93
		⁶⁰ Co	1.03	0.96	6.98	0.98	5.41
			1.04	0.95	8.41	0.98	6.00
Coffee	0.365	¹³⁷ Cs	4.09	4.76	16.45	3.94	3.55
		⁶⁰ Co	6.61	6.87	3.99	6.32	4.33
			6.61	6.70	1.34	6.72	1.60
Tea	0.453	¹³⁷ Cs	3.33	3.83	15.10	3.21	3.63
		⁶⁰ Co	5.38	5.51	2.46	5.16	4.08
			5.38	5.41	0.65	5.08	5.47
Unknown Syrup	1.15	¹³⁷ Cs	0.18	0.16	10.68	0.19	1.68
		⁶⁰ Co	0.37	0.35	5.26	0.35	7.01
Honey	1.363	¹³⁷ Cs	0.48	0.47	1.67	0.47	1.91
		⁶⁰ Co	0.77	0.70	10.08	0.74	4.20
			0.78	0.70	9.42	0.74	3.94



Conclusions

Computational Methods effective at predicting efficiencies over a range of densities

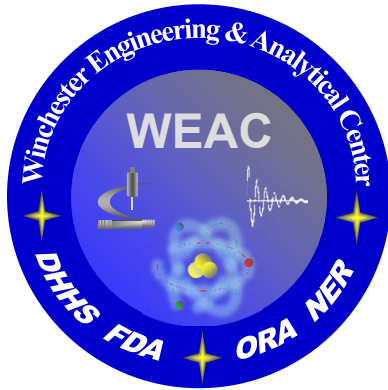
-LABSOCS

Organic Solutions can mimic food components, and allows for variation in density

Simulating radioactive solutions can help minimize generating waste saving cost of preparing samples and improve response time in the case of an emergency.



Acknowledgements

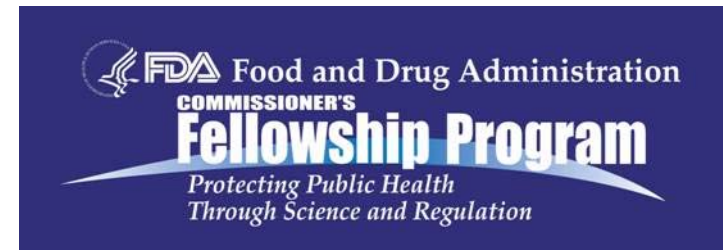


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THANK YOU!

Questions?



The Need for Food Density Corrections

Self absorption of gammas from radionuclides in a sample can affect the efficiency curve used to quantify sample activity.

The amount of correction can be trivial for isotopes with high energy photons (e.g. Cs-137) and favorable geometries (Marinelli Beakers).

Alternately, isotopes with significant low energy photon emissions (e.g. Am-241), and unfavorable counting geometries, such as large buckets of the food product, can require significant corrections.